

Mathematics for Chemistry

Wikibooks Forward

This book was created by volunteers at Wikibooks (<http://en.wikibooks.org>).

What is Wikibooks?

Started in 2003 as an offshoot of the popular Wikipedia project, Wikibooks is a free, collaborative wiki website dedicated to creating high-quality textbooks and other educational books for students around the world. In addition to English, Wikibooks is available in over 130 languages, a complete listing of which can be found at <http://www.wikibooks.org>. Wikibooks is a "wiki", which means anybody can edit the content there at any time. If you find an error or omission in this book, you can log on to Wikibooks to make corrections and additions as necessary. All of your changes go live on the website immediately, so your effort can be enjoyed and utilized by other readers and editors without delay.

Books at Wikibooks are written by volunteers, and can be accessed and printed for free from the website. Wikibooks is operated entirely by donations, and a certain portion of proceeds from sales is returned to the Wikimedia Foundation to help keep Wikibooks running smoothly. Because of the low overhead, we are able to produce and sell books for much cheaper than proprietary textbook publishers can. **This book can be edited by anybody at any time, including you.** We don't make you wait two years to get a new edition, and we don't stop selling old versions when a new one comes out.



Wikibooks is not a publisher of books, and is not responsible for the contributions of its volunteer editors. PediaPress.com is a print-on-demand publisher that is also not responsible for the content that it prints. Please see our disclaimer for more information:

[1]

What is this book?

This book was generated by the volunteers at Wikibooks, a team of people from around the world with varying backgrounds. The people who wrote this book may not be experts in the field. Some may not even have a passing familiarity with it. The result of this is that some information in this book may be incorrect, out of place, or misleading. For this reason, you should never rely on a community-edited Wikibook when dealing in matters of medical, legal, financial, or other importance. Please see our disclaimer for more details on this.

Despite the warning of the last paragraph, however, books at Wikibooks are continuously edited and improved. If errors are found they can be corrected immediately. If you find a problem in one of our books, we ask that you **be bold** in fixing it. You don't need anybody's permission to help or to make our books better.

Wikibooks runs off the assumption that many eyes can find many errors, and many able hands can fix them. Over time, with enough community involvement, the books at Wikibooks will become very high-quality indeed. **You are invited to participate at Wikibooks to help make our books better.** As you find problems in your book don't just complain about them: Log on and fix them! This is a kind of proactive and interactive reading experience that you probably aren't familiar with yet, so log on to <http://en.wikibooks.org> and take a look around at all the possibilities. We promise that we won't

bite!

Who are the authors?

The volunteers at Wikibooks come from around the world and have a wide range of educational and professional backgrounds. They come to Wikibooks for different reasons, and perform different tasks. Some Wikibookians are prolific authors, some are perceptive editors, some fancy illustrators, others diligent organizers. Some Wikibookians find and remove spam, vandalism, and other nonsense as it appears. Most wikibookians perform a combination of these jobs.

It's difficult to say who are the authors for any particular book, because so many hands have touched it and so many changes have been made over time. It's not unheard of for a book to have been edited thousands of times by hundreds of authors and editors. *You could be one of them too*, if you're interested in helping out. At the time this book was prepared for print, there have been over '**edits made by over 0**' registered users. These numbers are growing all the time.

Wikibooks in Class

Books at Wikibooks are free, and with the proper editing and preparation they can be used as cost-effective textbooks in the classroom or for independent learners. In addition to using a Wikibook as a traditional read-only learning aide, it can also become an interactive class project. Several classes have come to Wikibooks to write new books and improve old books as an interesting interactive class project. In some cases, the books written by students one year are used to teach students the next.

Happy Reading!

We at Wikibooks have put a lot of effort into these books, and we hope that you enjoy reading and learning from them. We want you to keep in mind that what you are holding is not a finished product but instead a work in progress. These books are never "finished" in the traditional sense, but they are ever-changing and evolving to meet the needs of readers and learners everywhere. Despite this constant change, we feel our books can be reliable and high-quality learning tools at a great price, and we hope you agree. Never hesitate to stop in at Wikibooks and make some edits of your own. We hope to see you there one day.

Happy reading!

External links

[1] http://en.wikibooks.org/wiki/Wikibooks:General_disclaimer

Source: http://en.wikibooks.org/w/index.php?title=Wikibooks:Collections_Preface&oldid=1332065

Principal Authors: Whiteknight, RobinH, Jomegat

Introduction

Foreword

This book was initially derived from a set of notes used in a university chemistry course. It is hoped it will evolve into something useful and develop a set of open access **problems** as well as pedagogical material.

For many universities the days when admission to a Chemistry, Chemical Engineering, Materials Science or even Physics course could require the equivalent of A-levels in Chemistry, Physics and Mathematics are probably over for ever. The broadening out of school curricula has had several effects, including student entry with a more diverse educational background and has also resulted in the subject areas Chemistry, Physics and Mathematics becoming **disjoint** so that there is no co-requisite material between them. This means that, for instance, physics cannot have any advanced, or even any very significant mathematics in it. This is to allow the subject to be studied without any of the maths which might be first studied by the A-level maths group at the ages of 17 and 18. Thus physics at school has become considerably more descriptive and visual than it was 20 years ago. The same applies to a lesser extent to chemistry.

This means there must be an essentially *remedial* component of university chemistry to teach just the Maths and Physics which is needed and not too much, if any more, as it is time consuming and perhaps not what the student of Chemistry is most focussed on. There is therefore also a need for a book Physics for chemistry.

Quantitative methods in chemistry

There are several reasons why numerical (quantitative) methods are useful in chemistry:

- Chemists need numerical information concerning reactions, such as how much of a substance is consumed, how long does this take, how likely is the reaction to take place.
- Chemists work with a variety of different units, with wildly different ranges, which one must be able to use and convert with ease.
- Measurements taken during experiments are not perfect, so evaluation and combination of errors is required.
- Predictions are often desired, and relationships represented as equations are manipulated and evaluated in order to obtain information.

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Introduction&oldid=1359549

Principal Authors: Amigadave

Numbers

Number theory

Numbers

See Number Theory for more information.

Real numbers come in several varieties and forms;

- Integers are whole numbers used for counting indivisible objects, together with negative equivalents and zero, *e.g.* 42, -7, 0
- Rational numbers can always be expressed as fractions, *e.g.* $4.673 = 4673/1000$.
- Irrational numbers, unlike rational numbers, cannot be expressed as a fraction or as a definite decimal, *e.g.* π and $\sqrt{2}$
- Natural numbers are integers that are greater than or equal to zero.

It is also worth noting that the imaginary unit and therefore complex numbers are used in chemistry, especially when dealing with equations concerning waves.

Surds

The origin of surds goes back to the Greek philosophers. It is relatively simple to prove that the square root of 2 cannot be a ratio of two integers, no matter how large the integers may become. In a rather Pythonesque incident the inventor of this proof was put to death for heresy by the other philosophers because they could not believe such a pure number as the root of 2 could have this *impure* property.

(The original use of quadratic equations is very old, Babylon many centuries BC.) This was to allocate land to farmers in the same quantity as traditionally held after the great floods on the Tigris and Euphrates had reshaped the fields. The mathematical technology became used for the same purpose in the Nile delta.

When you do trigonometry later you will see that surds are in the trigonometric functions of the important symmetrical angles, *e.g.* $\sin 60 = \frac{\sqrt{3}}{2}$ and so they appear frequently in mathematical expressions regarding 3 dimensional space.

Notation

The notation used for recording numbers in chemistry is the same as for other scientific disciplines, and appropriately called scientific notation, or *standard form*. It is a way of writing both very large and very small numbers in a shortened form compared to decimal notation. An example of a number written in scientific notation is

$$4.65 \times 10^6$$

with 4.65 being a coefficient termed the significand or the *mantissa*, and 6 being an integer exponent. When written in decimal notation, the number becomes

4650000 .

Numbers written in scientific notation are usually normalised, such that only one digit precedes the decimal point. This is to make order of magnitude comparisons easier, by simply comparing the exponents of two numbers written in scientific notation, but also to minimise transcription errors, as the decimal point has an assumed position after the first digit. In computing and on calculators, it is common for the $\times 10$ ("times ten to the power of") to be replaced with "E" (capital e). It is important not to confuse this "E" with the mathematical constant "e".

Engineering notation is a special restriction of scientific notation where the exponent must be divisible by three. Therefore, engineering notation is not normalised, but can easily use SI prefixes for magnitude.

Remember that in SI, numbers do not have commas between the thousands, instead there are spaces, *e.g.* 18 617 132 , (an integer) or 1.861 713 2 $\times 10^7$. Commas are used as decimal points in many countries.

Exponents

Order of calculations

Partial fractions

Partial fractions are used in a few derivations in thermodynamics and they are good for practicing algebra and factorisation.

It is possible to express quotients in more than one way. Of practical use is that they can be collected into one term or generated as several terms by the method of partial fractions. Integration of a complex single term quotient is often difficult, whereas by splitting it up into a sum, a sum of standard integrals is obtained. This is the principal chemical application of partial fractions.

An example is

$$\frac{x-2}{(x+1)(x-1)} = \frac{A(x-1) + B(x+1)}{(x+1)(x-1)} = \frac{A}{(x+1)} + \frac{B}{(x-1)}$$

In the above $x-2$ must equal $A(x-1) + B(x+1)$ since the denominators are equal. So we set x first to $+1$ giving $-1 = 2B$. Therefore $B = -1/2$. If we set $x = -1$ instead $-3 = -2A$, therefore $A = 3/2$. So

$$\frac{x-2}{(x+1)(x-1)} = \frac{3}{2(x+1)} - \frac{1}{2(x-1)}$$

We can reverse this process by use of a *common denominator*.

$$\frac{3}{2(x+1)} - \frac{1}{2(x-1)} = \frac{3(x-1) - (x+1)}{2(x+1)(x-1)}$$

The numerator is $2x-4$, so it becomes

$$\frac{(x-2)}{(x+1)(x-1)}$$

which is what we started from.

So we can generate a single term by multiplying by the denominators to create a common denominator and then add up the numerator to simplify. A typical application might be to convert a term to partial fractions, do some calculus on the terms, and then regather into

one quotient for display purposes. In a factorised single quotient it will be easier to see where numerators go to zero, giving solutions to $f(x) = 0$, and where denominators go to zero giving infinities.

A typical example of a meaningful infinity in chemistry might be an expression such as

$$\frac{A}{(E - E_a)^2}$$

The variable is the energy E , so this function is small everywhere, except near E_a . Near E_a a *resonance* occurs and the expression becomes infinite when the two energies are precisely the same. A molecule which can be electronically excited by light has several of these resonances.

Here is another example. If we had to integrate the following expression we would first convert to partial fractions:

$$\frac{3x}{2x^2 - 2x - 4} = \frac{3x}{2(x+1)(x-2)} = \frac{A}{x+1} + \frac{B}{x-2}$$

so

$$\frac{3}{2}x = A(x-2) + B(x-1)$$

let $x = 2$ then $3 = B$

let $x = 1$ then $\frac{3}{2} = -A$

therefore the expression becomes

$$\frac{3}{x-2} - \frac{3}{2(x+1)}$$

Later you will learn that these expressions integrate to give simple \ln expressions.

Problems

$$\begin{array}{ll} (1) & \frac{3}{x^2 - 1} \\ (2) & \frac{4x - 2}{x^2 + 2x} \\ (3) & \frac{4}{(2x + 1)(x - 3)} \\ (4) & \frac{7x + 6}{x^2 + 3x + 2} \\ (5) & \frac{x + 1}{2x^2 - 6x + 4} \end{array}$$

Answers

$$\begin{array}{ll} (1) & \frac{3}{2(x-1)} - \frac{3}{2(x+1)} \\ (2) & \frac{5}{(x+2)} - \frac{1}{x} \\ (3) & \frac{4}{7(x-3)} - \frac{8}{7(2x+1)} \\ (4) & \frac{8}{x+2} - \frac{1}{x+1} \\ (5) & \frac{3}{2(x-2)} - \frac{2}{2(x-1)} \end{array}$$

Polynomial division

This is related to partial fractions in that its principal use is to facilitate integration.

Divide out

$$\frac{3x^2 - 4x - 6}{1 + x}$$

like this

$$\begin{array}{r}
 3x \quad - 7 \\
 \hline
 x + 1 \quad) \quad 3x^2 \quad -4x \quad -6 \\
 \quad \quad 3x^2 \quad +3x \\
 \quad \quad \hline
 \quad \quad 0 \quad -7x \quad -6 \\
 \quad \quad \quad -7x \quad -7 \\
 \quad \quad \quad \hline
 \quad \quad \quad \quad \quad 1
 \end{array}$$

So our equation becomes $3x - 7 + \frac{1}{1 + x}$

This can be easily differentiated, and integrated. If this is differentiated with the quotient formula it is considerably harder to reduce to the the same form. The same procedure can be applied to partial fractions.

Substitutions and expansions

You can see the value of changing the variable by simplifying

$$\frac{\left(\frac{J(J+1)}{\epsilon}\right)^3 + 2\left(\frac{J(J+1)}{\epsilon}\right) + 1}{\left(\frac{J(J+1)}{\epsilon}\right)^2 - 9} - \frac{\left(\frac{J(J+1)}{\epsilon}\right)^2 - 2\left(\frac{J(J+1)}{\epsilon}\right) + 1}{\left(\frac{J(J+1)}{\epsilon}\right)^2 + 9}$$

to

$$\frac{x^3 + 2x + 1}{x^2 - 9} - \frac{x^2 - 2x + 1}{x^2 + 9}$$

where

$$x = \frac{J(J + 1)}{\epsilon}$$

This is an example of simplification. It would actually be possible to differentiate this with respect to either J or ϵ using only the techniques you have been shown. The algebraic manipulation involves *differentiation of a quotient* and the *chain rule*.

Evaluating $\frac{dy}{dx}$ gives

$$\frac{3x^2 + 2}{x^2 - 9} - 2 \frac{(x^3 + 2x + 1)x}{(x^2 - 9)^2} - \frac{2x - 2}{x^2 + 9} + 2 \frac{(x^2 - 2x + 1)x}{(x^2 + 9)^2}$$

Expanding this out to the J s and ϵ s would look ridiculous.

Substitutions like this are continually made for the purpose of having new, simpler expressions, to which the rules of calculus or identities are applied.

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Number_theory&oldid=1359599

Principal Authors: Amigadave, MartinY, Jguk, Whiteknight

Functions

This chapter is a 'stub'. You can help Wikibooks by reorganizing or for chemistry/Functions expanding it ^[1].

Functions as tools in chemistry

The quadratic formula

In order to find the solutions to the general form of a quadratic equation

$$ax^2 + bx + c$$

there is a formula

$$x = \frac{-b \pm \sqrt{(b^2 - 4ac)}}{2a}$$

(Notice the line over the square root has the same *priority* as a bracket. Of course we all know by now that $\sqrt{a+b}$ is not equal to $\sqrt{a} + \sqrt{b}$ but errors of priority are among the most common algebra errors in practice).

There is a formula for a cubic equation but it is rather complicated and unlikely to be required for undergraduate-level study of chemistry. Cubic and higher equations occur often in chemistry but if they do not factorise they are usually solved by computer.

Solve:

$$2x^2 - 14x + 9$$

$$1.56(x^2 + 3.67x + 0.014)$$

Notice the *scope* or *range* of the bracket.

$$2x^2 - 4x + 2$$

$$-45.1(1.2[A]^2 - 57.9[A] + 4.193)$$

Notice here that the variable is a *concentration*, not the ubiquitous x .

External links

[1] <http://en.wikipedia.org/wiki/:Mathematics>

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Functions&oldid=1359565

Principal Authors: Amigadave

Units and dimensions

Units, multipliers and prefixes

It is usually necessary in chemistry to be familiar with at least three systems of units, Le Système International d'Unités (SI), atomic units as used in theoretical calculations and the unit system used by the experimentalists. Thus if we are dealing with the ionization energy, the units involved will be the Joule (J), the Hartree (E_h , the atomic unit of energy), and the electron volt (eV).

These units all have their own advantages;

- The SI unit should be understood by all scientists regardless of their field.
- The atomic unit system is the natural unit for theory as most of the fundamental constants are unity and equations can be cast in dimensionless forms.
- The electron volt comes from the operation of the ionization apparatus where individual electrons are accelerated between plates which have a potential difference in Volts.

An advantage of the SI system is that the dimensionality of each term is made clear as the fundamental constants have structure. This is a complicated way of saying that if you know the dimensionality of all the things you are working with you know an awful lot about the mathematics and properties such as scaling with size of your system. Also, the same system of units can describe both the output of a large power station (gigaJoules), or the interaction of two inert gas atoms, (a few kJ per mole or a very small number of Joules per molecule when it has been divided by Avogadro's number).

In SI the symbols for units are lower case unless derived from a person's name, *e.g.* ampere is A and kelvin is K.

SI base units

Name	Symbol	Quantity
metre	m	length
kilogram	kg	mass
second	s	time
ampere	A	electric current
kelvin	K	thermodynamic temperature
candela	cd	luminous intensity
mole	mol	amount of substance

Derived units used in chemistry

Quantity	Unit	Name	Symbol
Area	m^2		
Volume	m^3		
Velocity	$m s^{-1}$		
Acceleration	$m s^{-2}$		
Density	$kg m^{-3}$		
Entropy	$J mol^{-1} K^{-1}$		

Force	kg m s^{-2}	newton	N
Energy	N m	joule	J
Pressure	N m^{-1}	pascal	Pa
Frequency	s^{-1}	hertz	Hz

Approved prefixes for SI units

Prefix	Factor	Symbol
atto	10^{-18}	a
femto	10^{-15}	f
pico	10^{-12}	p
nano	10^{-9}	n
micro	10^{-6}	μ
milli	10^{-3}	m
centi	10^{-2}	c
deci	10^{-1}	d
kilo	10^3	k
mega	10^6	M
giga	10^9	G
tera	10^{12}	T
peta	10^{15}	P
exa	10^{18}	E

Note the use of capitals and lower case and the increment on the exponent being factors of 3. Notice also centi and deci are supposed to disappear with time leaving only the powers of 1000.

Conversion factors

The $\hat{}$, (sometimes call caret or hat), sign is another notation for *to the power of*. E means *times 10 to the power of*, and is used a great deal in computer program output.

Energy

An approximation of how much of a chemical bond each energy corresponds to is placed next to each one. This indicates that light of energy 4 eV can break chemical bonds and possibly be dangerous to life, whereas infrared radiation of a few cm^{-1} is harmless.

- 1 eV = 96.48530891 kJ mol⁻¹ (Near infrared), approximately 0.26 chemical bonds
- 1 kcal mol⁻¹ = 4.184000000 kJ mol⁻¹ (Near infrared), approximately 0.01 chemical bonds
- 1 MHz = 0.399031E-06 kJ mol⁻¹ (Radio waves), approximately 0.00 chemical bonds
- 1 cm^{-1} = 0.01196265819 kJ mol⁻¹ (Far infrared), approximately 0.00 chemical bonds

Wavelength, generally measure in nanometres and used in UV spectroscopy is defined as an inverse and so has a reciprocal relationship.

Length

There is the metre, the Angstrom (10^{-10} m), the micron (10^{-6} m), the astronomical unit (AU) and many old units such as feet, inches and light years.

Angles

The radian to degree conversion is 57.2957795130824, (i.e. a little bit less than 60, remember your equilateral triangle and radian sector).

Dipole moment

1 Debye = $3.335640035 \text{ Cm} \times 10^{-30}$ (coulomb metre)

Magnetic Susceptibility

$1 \text{ cm}^3 \text{ mol}^{-1} = 16.60540984 \text{ JT}^{-2} \times 10^{30}$ (joule tesla²)

Old units

Occasionally, knowledge of older units may be required. Imperial units, or convert energies from BTUs in a thermodynamics project, etc.

In university laboratory classes you would probably be given material on the *Quantity Calculus* notation and methodology which is to be highly recommended for scientific work.

A good reference for units, quantity calculus and SI is: I. Mills, T. Cuitas, K. Homann, N. Kallay, K. Kuchitsu, *Quantities, units and symbols in physical chemistry, 2nd edition*, (Oxford:Blackwell Scientific Publications,1993).

Greek alphabet

Unit labels

The labelling of tables and axes of graphs should be done so that the numbers are dimensionless, *e.g.* temperature is T/K ,

$\ln(k/k_0)$ and energy mol / kJ *etc.*

This can look a little strange at first. Examine good text books like Atkins Physical Chemistry which follow SI carefully to see this in action.

The hardest thing with conversion factors is to get them the right way round. A common error is to divide when you should be multiplying, also another common error is to fail to raise a conversion factor to a power.

$$\text{Conversion factor} = \frac{\text{Units required}}{\text{Units given}}$$

$$1 \text{ eV} = 96.48530891 \text{ kJ mol}^{-1}$$

$$1 \text{ cm}^{-1} = 0.01196265819 \text{ kJ mol}^{-1}$$

To convert eV to cm^{-1} , first convert to kJ per mole by multiplying by 96.48530891 / 1. Then convert to cm^{-1} by multiplying by 1 / 0.01196265819 giving 8065.540901. If we tried to go directly to the conversion factor in 1 step it is easy to get it upside down. However, common sense tells us that there are a lot of cm^{-1} s in an eV so it should be obviously wrong.

1 inch = 2.54 centimetres. If there is a surface of nickel electrode of $2 * 1.5$ square inches it must be $2 * 1.5 * 2.54^2$ square centimetres.

To convert to square metres, the SI unit we must divide this by $100 * 100$ not just 100.

Dimensional analysis

The technique of adding unit labels to numbers is especially useful, in that analysis of the units in an equation can be used to double-check the answer.

An aside on scaling

One of the reasons powers of variables are so important is because they relate to the way quantities scale. Physicists in particular are interested in the way variables scale in the limit of very large values. Take cooking the turkey for Christmas dinner. The amount of turkey you can afford is linear, (power 1), in your income. The size of an individual serving is quadratic, (power 2), in the radius of the plates being used. The cooking time will be something like cubic in the diameter of the turkey as it can be presumed to be linear in the mass.

(In the limit of a very large turkey, say one the diameter of the earth being heated up by a nearby star, the internal conductivity of the turkey would limit the cooking time and the time taken would be exponential. No power can go faster / steeper than exponential in the limit. The series expansion of e^x goes on forever even though $1/n!$ gets very small.)

Another example of this is why dinosaurs had fatter legs than modern lizards. If dinosaurs had legs in proportion to small lizards the mass to be supported rises as length to the power 3 but the strength of the legs only rises as the area of the cross section, power 2. Therefore the bigger the animal the more enormous the legs must become, which is why a rhino is a very chunky looking version of a pig.

There is a very good article on this in Cooper, Necia Grant; West, Geoffrey B., *Particle Physics: A Los Alamos Primer*, ISBN 0521347807

Making tables

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Units_and_dimensions&oldid=1359572

Principal Authors: MartinY, Amigadave, Jguk, Whiteknight

Statistics

This chapter is a 'stub. You can help Wikibooks by reorganizing or for chemistry/Statistics expanding it ^[1].

Definition of errors

For a quantity x the error is defined as Δx . Consider a burette which can be read to $\pm 0.05 \text{ cm}^3$ where the volume is measured at 50 cm^3 .

- The *absolute error* is $\pm \Delta x, \pm 0.05 \text{ cm}^3$
- The *fractional error* is $\pm \frac{\Delta x}{x}, \pm \frac{0.05}{50} = \pm 0.001$
- The *percentage error* is $\pm 100 \times \frac{\Delta x}{x} = \pm 0.1 \%$

Combination of uncertainties

In an experimental situation, values with errors are often combined to give a resultant value. Therefore, it is necessary to understand how to combine the errors at each stage of the calculation.

Addition or subtraction

Assuming that Δx and Δy are the errors in measuring x and y , and that the two variables are combined by addition or subtraction, the uncertainty (absolute error) may be obtained by calculating

$$\sqrt{(\Delta x)^2 + (\Delta y)^2}$$

which can be expressed as a relative or percentage error if necessary.

Multiplication or division

Assuming that Δx and Δy are the errors in measuring x and y , and that the two variables are combined by multiplication or division, the fractional error may be obtained by calculating

$$\sqrt{\left(\frac{\Delta x}{x}\right)^2 + \left(\frac{\Delta y}{y}\right)^2}$$

External links

[1] <http://en.wikipedia.org/wiki/:Mathematics>

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Statistics&oldid=1359517

Principal Authors: Amigadave, MartinY, Whiteknight

Plotting Graphs

The properties of graphs

The most basic relationship between two variables x and y is a straight line, a *linear relationship*.

$$y = mx + c$$

The variable m is the gradient and c is a constant which gives the intercept. The equations can be more complex than this including higher powers of x , such as

$$y = ax^2 + bx + c$$

This is called a *quadratic equation* and it follows a shape called a parabola. High powers of x can occur giving *cubic*, *quartic* and *quintic* equations. In general, as the power is increased, the line mapping the variables wiggles more, often cutting the x -axis several times.

Practice

Plot $x^2 - 1$ between -3 and +2 in units of 1.

Plot $x^2 + 3x$ between -4 and +1 in units of 1.

Plot $2x^3 - 5x^2 - 12x$ between -5 and +4 in units of 1.

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Plotting_graphs&oldid=1359533

Principal Authors: Amigadave

Complex numbers

Introduction to complex numbers

The equation:

$$x^2 + 6x + 13 = 0$$

Does not factorise

$$x = -3 \pm \sqrt{9 - 13}$$

$\sqrt{-4}$ without complex numbers does not exist. However the number $i = \sqrt{-1}$ behaves exactly like any other number in algebra without any anomalies, allowing us to solve this problem.

The solutions are $-3 \pm 2i$.

$2i$ is an *imaginary* number. $-3 - 2i$ is a *complex* number.

Two complex numbers $(a, b) = a + ib$ are added by $(a, b) + (c, d) = (a + c, b + d) = a + c + i(b + d)$.

Subtraction is obvious: $(a, b) - (c, d) = (a - c, b - d)$.

$$(a, b) \cdot (c, d) = ac + i(ad + bc) - bd$$

Division can be worked out as an exercise. It requires $(c + id)(c - id)$ as a common denominator. This is $c^2 - (id)^2$, (difference of two squares), and is $c^2 + d^2$.

This means

$$\frac{(a, b)}{(c, d)} = \frac{ac + bd + i(cb - ad)}{c^2 + d^2}$$

In practice complex numbers allow one to simplify the mathematics of magnetism and angular momentum as well as completing the number system.

There is an apparent one to one correspondence between the Cartesian $x - y$ plane and the complex numbers, $x + iy$. This is called an Argand diagram. The correspondence is illusory however, because say for example you raise the square root of i to a series of ascending powers. Rather than getting larger it goes round and round in circles around the origin. This is not a property of ordinary numbers and is one of the fundamental features of behaviour in the complex plane.

Plot on the same Argand diagram $2 - i$, $3i - 1$, $-2 - 2i$

Solve

$$x^2 + 4x + 29 = 0$$

$$4x^2 - 12x + 25 = 0$$

$$x^2 + 2ix + 1 = 0$$

(Answers -2 plus or minus $5i$, $3/2$ plus or minus $2i$, $i(-1$ plus or minus root $2)$)

2 important equations to be familiar with, Euler's equation:

$$e^{i\theta} = \cos \theta + i \sin \theta$$

and de Moivre's theorem:

$$(\cos \theta + i \sin \theta)^n = \cos n\theta + i \sin n\theta$$

Euler's equation is obvious from looking at the Maclaurin expansion of $e^{i\theta}$.

To find the square root of i we use de Moivre's theorem.

$$e^{i\frac{\pi}{2}} = 0 + 1i$$

so de Moivre's theorem gives $\sqrt{e^{i\frac{\pi}{2}}} = \cos \frac{\pi}{2.2} + i \sin \frac{\pi}{2.2}$

$$= \frac{1}{\sqrt{2}} + \frac{1}{\sqrt{2}}i$$

Check this by squaring up to give i .

The other root comes from:

$$\frac{5\pi}{4} + i \sin \frac{5\pi}{4} = -\frac{1}{\sqrt{2}} - \frac{1}{\sqrt{2}}i$$

de Moivre's theorem can be used to find the *three* cube roots of unity by

$$1 = \cos \theta + i \sin \theta$$

where θ can be $0 \pm 2\pi/n$.

Put $n = 1/3$, $\cos \theta = -1/2$ and $\sin \theta = \pm\sqrt{3}/2$

$$\begin{aligned} \left(-\frac{1}{2} + \frac{\sqrt{3}}{2}i\right)^3 &= \left(\frac{1}{4} - \frac{3}{4} - \frac{\sqrt{3}}{2}i\right) \left(-\frac{1}{2} + \frac{\sqrt{3}}{2}i\right) \\ &= \left(-\frac{1}{2} - \frac{\sqrt{3}}{2}i\right) \left(-\frac{1}{2} + \frac{\sqrt{3}}{2}i\right) \end{aligned}$$

This is the difference of two squares so

$$\left(\frac{1}{2}\right)^2 - \frac{3}{4}i^2 = \frac{1}{4} + \frac{3}{4}$$

Similarly any collection of n th roots of 1 can be obtained this way.

Another example is to get the expressions for $\cos 4\theta$ and $\sin 4\theta$ without expanding $\cos(2\theta + 2\theta)$.

$$\cos 4\theta + i \sin 4\theta = (\cos \theta + i \sin \theta)^4$$

Remember Pascal's Triangle

			1			
		1	2	1		
	1	3	3	1		
	1	4	6	4	1	
	1	5	10	10	5	1
1	6	15	20	15	6	1

$$= \cos^4 \theta + 4 \cos^3 \theta i \sin \theta + 6 \cos^2 \theta i^2 \sin^2 \theta + 4 \cos \theta i^3 \sin^3 \theta + i^4 \sin^4 \theta$$

$$= \cos^4 \theta - 6 \cos^2 \theta \sin^2 \theta + \sin^4 \theta + i(4 \cos^3 \theta \sin \theta - 4 \cos \theta \sin^3 \theta)$$

Separating the real and imaginary parts gives the two expressions. This is a lot easier than

$$\cos(2\theta + 2\theta)$$

Use the same procedure to get

$$\cos 6\theta \text{ and } \sin 6\theta .$$

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Complex_Numbers&oldid=1359388

Principal Authors: MartinY, Amigadave, Jguk, Whiteknight

Trigonometry, matrices

Trigonometry

Free Web Based Material from UK HEFCE

There is a DVD on trigonometry at Math Tutor^[1].

Trigonometry -the sin and cosine rules

frame right Trigonometric triangle

In the following trigonometric identities, a , b and c are the lengths of sides in a triangle, opposite the corresponding angles A , B and C .

- The sin rule $\frac{a}{\sin A} = \frac{b}{\sin B} = \frac{c}{\sin C}$
- The cosine rule $c^2 = a^2 + b^2 - 2ab \cos C$
- The ratio identity $\tan A = \frac{\sin A}{\cos A}$

Trigonometric identities

$$\cos^2 \theta + \sin^2 \theta = 1$$

Remember this is a consequence of Pythagoras' theorem where the length of the hypotenuse is 1.

$$\sin(\theta + \phi) = \sin \theta \cos \phi + \cos \theta \sin \phi$$

$$\cos(\theta + \phi) = \cos \theta \cos \phi - \sin \theta \sin \phi$$

The difference of two angles can easily be generated by putting $\phi = -\phi$ and remembering $\sin -\phi = -\sin \phi$ and $\cos -\phi = \cos \phi$. Similarly, the double angle formulae are generated by induction. $\tan(\theta + \phi)$ is a little more complicated but can be generated if you can handle the fractions! The proofs are in many textbooks but as a chemist it is not necessary to know them, only the results.

Identities and equations

Identities and equations look very similar, two things connected by an equals sign. An identity however is a memory aid of a mathematical equivalence and can be proved. An equation represents new information about a situation and can be solved.

For instance,

$$\cos^2 \theta + \sin^2 \theta = 1$$

is an identity. It cannot be solved for θ . It is valid for all θ . However,

$$\cos^2 \theta = 1$$

is an equation where $\theta = \arccos \pm 1$.

If you try and solve an identity as an equation you will go round and round in circles getting nowhere, but it would be possible to dress up $\cos^2 \theta + \sin^2 \theta = 1$ into a very complicated

expression which you could mistake for an equation.

Some observations on triangles

Check you are familiar with your elementary geometry. Remember from your GCSE maths the properties of equilateral and isosceles triangles. If you have an isosceles triangle you can always dispense with the sin and cosine rules, drop a perpendicular down to the base and use trig directly. Remember that the bisector of a side or an angle to a vertex cuts the triangle in two by area, angle and length. This can be demonstrated by drawing an obtuse triangle and seeing that the areas are $\frac{1}{2} \cdot R \cdot h$.

The interior angles of a polygon

Remember that the interior angles of a n -sided polygon are $n * 180 - 360$,

($n\pi - 2\pi$).

For benzene there are six *equilateral* triangles if the centre of the ring is used as a vertex, each of which having an interior angle of 120 degrees. Work out the angles in azulene, (a hydrocarbon with a five and a seven membered ring), assuming all the C-C bond lengths are equal, (this is only approximately true).

External links

[1] <http://www.mathtutor.ac.uk/viewdisksinfo.php?disk=5>

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Trigonometry&oldid=1359371

Principal Authors: MartinY, Amigadave, Whiteknight

Vectors

Free Web Based Material from HEFCE

There is a DVD on vectors at Math Tutor ^[1].

Vectors

Imagine you make a rail journey from Doncaster to Bristol, from where you travel up the West of the country to Manchester. Here you stay a day, travelling the next morning to Glasgow, then across to Edinburgh. At the end of a day's work you return to Doncaster. Mathematically this journey could be represented by vectors, (in 2 dimensions because we are flat earthers on this scale). At the end of the 2nd journey (D-B) + (B-M) you are only a short distance from Doncaster, 50 miles at 9.15 on the clockface. Adding two more vectors, (journeys) takes you to Edinburgh, (about 250 miles at 12.00). Completing the journey leaves you at a zero vector away from Doncaster, *i.e.* all the vectors in this closed path add to zero.

Mathematically we usually use 3 dimensional vectors over the 3 Cartesian axes x , y and z .

It is best always to use the conventional right handed axes even though the other way round is equally valid if used consistently. The wrong handed coordinates can occasionally be found erroneously in published research papers and text books. The memory trick is to think of a sheet of graph paper, x is across as usual and y up the paper. Positive z then comes out of the paper.

A *unit vector* is a vector *normalised*, i.e. multiplied by a constant so that its value is 1. We have the unit vectors in the 3 dimensions:

$$\hat{i} + \hat{j} + \hat{k}$$

so that

$$\mathbf{v} = A_x \hat{i} + A_y \hat{j} + A_z \hat{k}$$

The hat on the i, j, k signifies that it is a *unit vector*. This is usually omitted.

Our geographical analogy allows us to see the meaning of vector addition and subtraction. Vector **products** are less obvious and there are two definitions the *scalar product* and the *vector product*. These are different kinds of mathematical animal and have very different applications. A scalar product is an area and is therefore an ordinary number, a scalar. This has many useful trigonometrical features.

The vector product seems at first to be defined rather strangely but this definition maps onto nature as a very elegant way of describing *angular momentum*. The structure of Maxwell's Equations is such that this definition simplifies all kinds of mathematical descriptions of atomic / molecular structure and electricity and magnetism.

A summary of vectors

The unit vectors in the 3 Cartesian dimensions:

$$\hat{i} + \hat{j} + \hat{k}$$

a vector \mathbf{v} is:

$$\mathbf{v} = A_x \hat{i} + A_y \hat{j} + A_z \hat{k}$$

The hat on the i, j, k signifies that it is a *unit vector*.

Vector magnitude

$$V_{mag} = \sqrt{A_x^2 + A_y^2 + A_z^2}$$

A constant times a vector

$$\mathbf{v}_{new} = cA_x \hat{i} + cA_y \hat{j} + cA_z \hat{k}$$

Vector addition

$$\mathbf{A} + \mathbf{B} = (A_x + B_x) \hat{i} + (A_y + B_y) \hat{j} + (A_z + B_z) \hat{k}$$

Notice $\mathbf{B} + \mathbf{A} = \mathbf{A} + \mathbf{B}$

Vector subtraction

$$\mathbf{A} - \mathbf{B} = (A_x - B_x) \hat{i} + (A_y - B_y) \hat{j} + (A_z - B_z) \hat{k}$$

Notice $\mathbf{A} - \mathbf{B} = -(-\mathbf{A} + \mathbf{B})$

Scalar Product

$$\mathbf{A} \cdot \mathbf{B} = A_x \cdot B_x + A_y \cdot B_y + A_z \cdot B_z$$

Notice $\mathbf{B} \cdot \mathbf{A} = \mathbf{A} \cdot \mathbf{B}$

Notice that if $\mathbf{A} = \mathbf{B}$ this reduces to a square.

If \mathbf{A} and \mathbf{B} have no common non-zero components in x , y and z the value is zero corresponding to *orthogonality*, i.e. they are at right angles. (This can also occur by sign combinations making $\mathbf{A} \cdot \mathbf{B}$ zero, corresponding to non axis-hugging right angles.)

Vector product

$$\mathbf{A} \times \mathbf{B} = (A_y B_z - A_z B_y) \hat{\mathbf{i}} - (A_x B_z - A_z B_x) \hat{\mathbf{j}} + (A_x B_y - A_y B_x) \hat{\mathbf{k}}$$

Notice $\mathbf{B} \times \mathbf{A} = -\mathbf{A} \times \mathbf{B}$

The minus sign on the middle term comes from the definition of the determinant, explained in the lecture. Determinants are defined that way so they correspond to right handed rotation. (If you remember our picture of $\cos^2 + \sin^2 = 1$ going round the circle, as one coordinate goes up, i.e. more positive, another must go down. Therefore rotation formulae must have both negative and positive terms.) Determinants are related to rotations and the solution of simultaneous equations. The solution of n simultaneous equations can be recast in graphical form as a rotation to a unit vector in n -dimensional space so therefore the same mathematical structures apply to both space and simultaneous equations.

External links

[1] <http://www.mathtutor.ac.uk/viewdisksinfo.php?disk=4>

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Vectors&oldid=1359405

Principal Authors: MartinY, Amigadave, Whiteknight

Matrices and determinants

Simultaneous linear equations

If we have 2 equations of the form $y = mx + c$ we may have a set of *simultaneous equations*. Suppose two rounds of drinks are bought in a cafe, one round is 4 halves of orange juice and 4 packets of crisps. This comes to 4 pounds 20. The thirstier drinkers at another table buy 4 pints of orange juice and only 1 packet of crisps and this comes to 6 pounds 30. So we have:

$$4.20 = 2x + 4y$$

and

$$6.30 = 4x + y$$

$$\text{i.e. } y = \frac{4.20}{4} - \frac{2}{4}x \quad \text{and} \quad y = 6.30 - 4x$$

If you plot these equations they will be simultaneously true at $x = 1.5$ and $y = 0.30$.

Notice that if the two rounds of drinks are 2 pints and 2 packets of crisps and 3 pints and 3 packets of crisps we cannot solve for the prices! This corresponds to two parallel straight lines which never intersect.

If we have the equations:

$$3x + 4y = 4$$

$$3x + 2y = 1$$

If these are simultaneously true we can find a *unique* solution for both x and y .

By subtracting the 2 equations a new equation is created where x has disappeared and the system is solved.

$$2y = 3$$

Substituting back $y = 1.5$ gives us x .

This was especially easy because x had the same coefficient in both equations. We can always multiply one equation throughout by a constant to make the coefficients the same.

If the equations were:

$$3x + 4y = 4$$

and

$$6x + 8y = 8$$

things would go horribly wrong when you tried to solve them because they are two copies of the same equation and therefore not *simultaneous*. We will come to this later, but in the meantime notice that 3 times 8 = 4 times 6. If our equations were:

$$3x + 4y + 9z = 4$$

$$3x + 2y - 4z = 1$$

$$9x + 2y - 2z = 1$$

we can still solve them but would require a lot of algebra to reduce it to three (2x2) problems which we know we can solve. This leads on to the subject of matrices and determinants.

Simultaneous equations have applications throughout the physical sciences and range in size from (2x2)s to sets of equations over 1 million by 1 million.

Practice simultaneous equations

Solve:

$$x - 2y = 1$$

$$x + 4y = 8$$

and

$$x + 5y = 10$$

$$7x - 4y = 10$$

Notice that you can solve:

$$3x + 4y + 9z = 4$$

$$3x + 2y - 4z = 1$$

$$0x + 0y - 2z = 1$$

because it breaks down into a (2x2) and is not truly a (3x3). (In the case of the benzene molecular orbitals, which are (6x6), this same scheme applies. It becomes two direct solutions and two (2x2) problems which can be solved as above.)

Matrices

The multiplication of matrices has been explained in the lecture.

$$A = \begin{pmatrix} 0 & 1 & 2 \\ 0 & -1 & -2 \end{pmatrix} \quad B = \begin{pmatrix} 3 & -3 \\ 4 & -4 \\ 5 & -5 \end{pmatrix} \quad AB = \begin{pmatrix} 14 & -14 \\ -14 & 14 \end{pmatrix}$$

$$BA = \begin{pmatrix} 0 & 6 & 12 \\ 0 & 8 & 16 \\ 0 & 10 & 20 \end{pmatrix}$$

$$\text{if } A = (1 \ 2 \ 3) \quad \text{and} \quad B = \begin{pmatrix} 4 & 5 \\ 6 & 7 \\ 8 & 9 \end{pmatrix} \quad AB = (40 \ 60)$$

but BA cannot exist. To be multiplied two matrices must have the 1st matrix with the same number of elements in a row as the 2nd matrix has elements in its columns.

$$a_{ij} = \sum_k a_{ik} b_{kj}$$

where the a_{ij} s are the elements of A .

$$\begin{pmatrix} a_{11} & a_{12} & a_{13} \\ a_{21} & a_{22} & a_{23} \\ a_{31} & a_{32} & a_{33} \end{pmatrix}$$

Look at our picture of \cos and \sin as represented by a unit vector in a circle. The rotation of the unit vector \hat{j} about the z -axis can be represented by the following mathematical construct.

$$\begin{pmatrix} \cos(\theta) & \sin(\theta) & 0 \\ -\sin(\theta) & \cos(\theta) & 0 \\ 0 & 0 & 1 \end{pmatrix} \begin{pmatrix} 0 \\ 1 \\ 0 \end{pmatrix} = \begin{pmatrix} \sin(\theta) \\ \cos(\theta) \\ 0 \end{pmatrix}$$

In two dimensions we will rotate the vector at 45 degrees between x and y :

$$\begin{pmatrix} \cos(\theta) & \sin(\theta) \\ -\sin(\theta) & \cos(\theta) \end{pmatrix} \begin{pmatrix} \frac{r}{\sqrt{2}} \\ \frac{r}{\sqrt{2}} \end{pmatrix} = \begin{pmatrix} \frac{r \cos \theta}{\sqrt{2}} + \frac{r \sin \theta}{\sqrt{2}} \\ -\frac{r \sin \theta}{\sqrt{2}} + \frac{r \cos \theta}{\sqrt{2}} \end{pmatrix} = \begin{pmatrix} r \\ 0 \end{pmatrix}$$

This is if we rotate by +45 degrees. For $\theta = -45^\circ$ $\cos(-45) = \cos 45$ and $\sin(-45) = -\sin 45$. So the rotation flips over to give (01) . The minus sign is necessary for the correct mathematics of rotation and is in the lower left element to give a right handed sense to the rotational sign convention.

As discussed earlier the solving of simultaneous equations is equivalent in some deeper sense to rotation in n -dimensions.

Matrix multiply practice

i) Multiply the following (2x2) matrices.

$$\begin{pmatrix} 3 & 5 \\ 6 & 4 \end{pmatrix} \begin{pmatrix} 1 & 2 \\ 3 & 4 \end{pmatrix} = \begin{pmatrix} 3x1 + 5x3 & 3x2 + 5x4 \\ 6x1 + 4x3 & 6x2 + 4x4 \end{pmatrix}$$

ii) Multiply the following (3x3) matrices.

$$\begin{pmatrix} 38.15 & -42.17 & 4.02 \\ 4.69 & 0.76 & -5.35 \\ -9.94 & 8.20 & 2.74 \end{pmatrix} \begin{pmatrix} 0.205 & 0.665 & 1 \\ 0.181 & 0.647 & 1 \\ 0.207 & 0.476 & 1 \end{pmatrix}$$

You will notice that this gives a *unit matrix* as its product.

$$\begin{pmatrix} 1 & 0 & 0 \\ 0 & 1 & 0 \\ 0 & 0 & 1 \end{pmatrix}$$

The first matrix is the *inverse* of the 2nd. Computers use the inverse of a matrix to solve simultaneous equations.

If we have

$$\begin{pmatrix} y_1 = a_{11}x_1 + a_{12}x_2 + a_{13}x_3 \dots + a_{1n}x_n \\ y_2 = a_{21}x_1 + a_{22}x_2 + a_{23}x_3 \dots + a_{2n}x_n \\ \dots \\ \dots \\ y_n = a_{n1}x_1 + a_{n2}x_2 + a_{n3}x_3 \dots + a_{nn}x_n \end{pmatrix}$$

In matrix form this is....

$$Y = AX$$

$$A^{-1}Y = X$$

In terms of work this is equivalent to the elimination method you have already employed for small equations but can be performed by computers for 10 000 simultaneous equations.

(Examples of large systems of equations are the fitting of reference data to 200 references molecules, dimension 200, or the calculation of the quantum mechanical gradient of the energy where there is an equation for every way of exciting 1 electron from an occupied orbital to an excited, (called *virtual*, orbital, (typically 10 000 equations.)

Finding the inverse

How do you find the inverse... You use Maple or Matlab on your PC but if the matrix is small you can use the formula...

$$A^{-1} = \frac{1}{\text{Det}A} \text{Adj}A$$

Here Adj A is the adjoint matrix, the transposed matrix of cofactors. These strange objects are best described by example.....

$$\begin{vmatrix} 1 & -1 & 2 \\ 2 & 1 & 1 \\ 3 & -1 & 1 \end{vmatrix}$$

This determinant is equal to: 1 (1 x 1 - 1 x (-1)) - (-1) (2 x 1 - 1 x 3) + 2 (2 x (-1) - (1 x 3) each of these terms is called a *cofactor*.

$$\begin{vmatrix} a_{11} & a_{12} & a_{13} \\ a_{21} & a_{22} & a_{23} \\ a_{31} & a_{32} & a_{33} \end{vmatrix}$$

$$\text{Det}A = -1^{i+j-1} \left(\begin{array}{l} a_{11}(a_{22}a_{33} - a_{23}a_{32}) \\ + -1^{i+j-1}a_{12}(a_{21}a_{33} - a_{23}a_{31}) \\ + -1^{i+j-1}a_{13}(a_{21}a_{32} - a_{22}a_{31}) \end{array} \right)$$

This -1^{i+j-1} thing gives the sign alternation in a form mathematicians like even though it is incomprehensible.

Use the determinant

$$\begin{vmatrix} 1 & -1 & 2 \\ 2 & 1 & 1 \\ 3 & -1 & 1 \end{vmatrix}$$

to solve the simultaneous equations on page 47 by the matrix inverse method. The matrix corresponding to the equations on p47.2 is:

$$\begin{array}{ccc} 1 & -1 & 2 & & 6 \\ 2 & 1 & 1 & = & 3 \\ 3 & -1 & 1 & & 6 \end{array}$$

The cofactors are

$$\begin{array}{ccc} 2 & 1 & -5 \\ -1 & -5 & -2 \\ -3 & 3 & 3 \end{array}$$

You may find these 9 copies of the matrix useful for striking out rows and columns to form this inverse....

$$\begin{array}{ccc} 1 & -1 & 2 & 1 & -1 & 2 & 1 & -1 & 2 \\ 2 & 1 & 1 & 2 & 1 & 1 & 2 & 1 & 1 \\ 3 & -1 & 1 & 3 & -1 & 1 & 3 & -1 & 1 \end{array}$$

$$\begin{array}{ccc} 1 & -1 & 2 & 1 & -1 & 2 & 1 & -1 & 2 \\ 2 & 1 & 1 & 2 & 1 & 1 & 2 & 1 & 1 \\ 3 & -1 & 1 & 3 & -1 & 1 & 3 & -1 & 1 \end{array}$$

$$\begin{array}{ccc} 1 & -1 & 2 & 1 & -1 & 2 & 1 & -1 & 2 \end{array}$$

$$\begin{array}{ccc|ccc|ccc} 2 & 1 & 1 & 2 & 1 & 1 & 2 & 1 & 1 \\ 3 & -1 & 1 & 3 & -1 & 1 & 3 & -1 & 1 \end{array}$$

These are the little determinants with the -1 to the (n-1) factors and the value of the determinant is -9.

The transposed matrix of cofactors is

$$\begin{array}{ccc} 2 & -1 & -3 \\ 1 & -5 & 3 \\ -5 & -2 & 3 \end{array}$$

So the inverse is

$$\begin{array}{ccc|ccc} & & & 2 & -1 & -3 \\ -1/9 \ X & & & 1 & -5 & 3 \\ & & & -5 & -2 & 3 \end{array}$$

Giving a solution

$$\begin{array}{ccc|ccc|cc} & & & 2 & -1 & -3 & 6 & 1 \\ -1/9 \ X & & & 1 & -5 & 3 & X \ 3 & = \ -1 \\ & & & -5 & -2 & 3 & 6 & 2 \end{array}$$

This takes a long time to get all the signs right. Elimination by subtracting equations is MUCH easier. However as the computer cannot make sign mistakes it is not a problem when done by computer program.

The following determinant corresponds to an equation which is repeated three times giving an unsolvable set of simultaneous equations.

$$\begin{vmatrix} 1 & 2 & 3 \\ -1 & -2 & -3 \\ 2 & 4 & 6 \end{vmatrix}$$

Matrix multiplication is not necessarily commutative, which in English means AB does not equal BA all the time. Multiplication may not even be possible in the case of rectangular rather than square matrices.

I will put a list of the properties and definitions of matrices in an appendix for reference through the later years of the course.

Determinants and the Eigenvalue problem

In 2nd year quantum chemistry you will come across this object:

$$\begin{vmatrix} \alpha - E & \beta & 0 & 0 \\ \beta & \alpha - E & \beta & 0 \\ 0 & \beta & \alpha - E & \beta \\ 0 & 0 & \beta & \alpha - E \end{vmatrix} = 0$$

You divide by β and set $(\alpha - E)/\beta$ to equal x to get:

$$\begin{vmatrix} x & 1 & 0 & 0 \\ 1 & x & 1 & 0 \\ 0 & 1 & x & 1 \\ 0 & 0 & 1 & x \end{vmatrix} = 0$$

Expand this out and factorise it into two quadratic equations to give:

$$(x^2 + x - 1)(x^2 - x - 1) = 0$$

which can be solved using $x = -b \pm \text{etc.}$

Simultaneous equations as linear algebra

The above determinant is a special case of simultaneous equations which occurs all the time in chemistry, physics and engineering which looks like this:

$$\begin{pmatrix} (a_{11} - \lambda)x_1 + a_{12}x_2 + a_{13}x_3 \dots + a_{1n}x_n = 0 \\ a_{21}x_2 + (a_{22} - \lambda)x_1 + a_{23}x_3 \dots + a_{2n}x_n = 0 \\ \dots \\ \dots \\ a_{1n}x_n + a_{2n}x_n + a_{n3}x_3 \dots + (a_{11} - \lambda)x_1 = 0 \end{pmatrix}$$

This equation in matrix form is $(\mathbf{A} - \lambda\mathbf{1})\mathbf{x} = 0$ and the solution is $\text{Det}(\mathbf{A} - \lambda\mathbf{1}) = 0$.

This is a *polynomial equation* like the quartic above. As you know polynomial equations have as many solutions as the highest power of x i.e. in this case n . These solutions can be *degenerate* for example the π orbitals in benzene are a degenerate pair because of the factorisation of the x^6 polynomial from the 6 Carbon-pz orbitals. In the 2nd year you may do a lab exercise where you make the benzene determinant and see that the polynomial is

$$(x^2 - 4)(x^2 - 1)(x^2 - 1) = 0$$

from which the 6 solutions and the orbital picture are immediately obvious.

The use of matrix equations to solve arbitrarily large problems leads to a field of mathematics called *linear algebra*.

Matrices with complex numbers in them

Work out the quadratic equation from the 3 determinants

$$\begin{vmatrix} x & i \\ -i & x \end{vmatrix} \quad \begin{vmatrix} x & 1 \\ 1 & x \end{vmatrix} \quad \begin{vmatrix} x & \frac{1}{\sqrt{2}} + \frac{i}{\sqrt{2}} \\ \frac{1}{\sqrt{2}} - \frac{i}{\sqrt{2}} & x \end{vmatrix}$$

They are all the same! This exemplifies a deeper property of matrices which we will ignore for now other than to say that complex numbers allow you to calculate the same thing in different ways as well as being the *only* neat way to formulate some problems.

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Matrices_and_Determinants&oldid=1359429

Principal Authors: MartinY, Amigadave, Jguk, Whiteknight

Calculus

Differentiation

Free web-based material from HEFCE

There is a DVD on differentiation at Math Tutor ^[1].

The basic polynomial

The most basic kind of differentiation is:

$$f(x) = x^n$$

$$f'(x) = nx^{n-1}$$

There are two simple rules:

1. The derivative of a function times a constant is just the same constant times the derivative.
2. The derivative of a sum of functions is just the sum of the two derivatives.

To get higher derivatives such as the *second derivative* keep applying the same rules.

One of the big uses of differentiation is to find the stationary points of functions, the maxima and minima. If the function is smooth, (unlike a saw-tooth), these are easily located by solving equations where the first derivative is zero.

The chain rule

$$\frac{dy}{dx} = \frac{dy}{dw} \cdot \frac{dw}{dx}$$

This is best illustrated by example: find $\frac{dy}{dx}$ given $y = (x^4 + 1)^9$

Let $y = u^9$ and $u = x^4 + 1$.

Now $\frac{dy}{du} = 9u^8$ and $\frac{du}{dx} = 4x^3$

So using the chain rule we have $9(x^4 + 1)^8 \cdot 4x^3 = 36x^3(x^4 + 1)^8$

Differentiating a product

$$\frac{d(u.v)}{dx} = v \frac{du}{dx} + u \frac{dv}{dx}$$

Notice when differentiating a product one generates two *terms*. (Terms are mathematical expression connected by a plus or minus.) An important point is that terms which represent physical quantities must have the same units and dimensions or must be pure dimensionless numbers. You cannot add 3 oranges to 2 pears to get 5 orangopears.

Integration by parts also generates an extra term each time it is applied.

Differentiating a quotient

You use this to differentiate \tan .

$$\frac{d}{dx} \left(\frac{u}{v} \right) = \frac{v \frac{du}{dx} - u \frac{dv}{dx}}{v^2}$$

Problems

Differentiate with respect to x

$$3x^2 - x(1+x)(1-x)$$

$$\text{Notice we have } (a^2 - b^2) \cdot 4x^7 - 3x^2$$

$$(3x+2)2 + e^x$$

$$8x^6 - 12\sqrt{x}$$

$$x(x+3)2$$

$$x^2(3x - (2+x)(2-x))$$

Evaluate the inner brackets first.

Evaluate

$$\frac{d}{dz} \left(e^z - \frac{z^9}{18} - \frac{3}{z^2} \right)$$

$$\frac{d}{dc} (\sqrt{c^5})$$

$$\frac{d}{d\omega} \left(\frac{1}{\omega} - \frac{1}{\omega^2} - \frac{1}{\omega^3} \right)$$

$$\frac{d}{d\phi} \left(e^\phi - \left(\frac{1}{\phi^2} + \frac{3}{2}\phi^2 \right) \right)$$

$$\frac{d}{dr} \left(e^{5r} - \left(\frac{a}{r^3} + \frac{b}{r^6} + \frac{c}{r^9} \right) \right)$$

a , b and c are constants. Differentiate with respect to r .

$$3re^{-3r}$$

$$\frac{d}{dx} (x^6 e^x)$$

Answers

$$3x^2 + 6x - 1$$

$$28x^6 - 6x$$

$$e^x + 18x + 12$$

$$48x^5 - \frac{6}{\sqrt{x}}$$

$$3x^2 + 12x + 9$$

$$4x^3 + 9x^2 - 8x$$

$$e^z + \frac{z^8}{2} + \frac{6}{z^3}$$

$$\frac{5}{2}c^{3/2} \text{ or } \frac{5}{2}\sqrt{c^3}$$

$$-\frac{1}{\omega^2} + \frac{2}{\omega^3} + \frac{3}{\omega^4}$$

$$e^\phi + \frac{2}{\phi^3} - 3\phi$$

$$5e^{5r} + \frac{3a}{r^4} + \frac{6b}{r^7} + \frac{9c}{r^{10}}$$

$$e^{-3r}(3 - 9r)$$

$$x^5 e^x (x + 6)$$

Harder differentiation problems

Differentiate with respect to x :

$$x^4(x + 9)^5$$

$$5x^2(x^2 - 7x + 9)^5$$

Differentiate with respect to r

$$(r^2 + 3r - 1)^3 e^{-4r}$$

Differentiate with respect to ω

$$\frac{1}{\omega^4} (\omega^2 - 3\omega - 19)$$

$$\frac{e^\omega}{\omega^4}$$

Evaluate

$$\frac{d}{dz} (e^{-4z} (z - 1^2))$$

$$\frac{d}{d\phi} \left(\phi^2 e^{-2\phi} \left(1 - \frac{1}{\phi^2} \right) \right)$$

Using differentiation to check turning points

$\frac{dy}{dx}$ is the tangent or gradient. At a minimum $\frac{dy}{dx}$ is zero. This is also true at a maximum or

an inflection point. The *second gradient* gives us the nature of the point. If $\frac{d^2y}{dx^2}$ is positive

the turning point is a minimum and if it is negative a maximum. Most of the time we are interested in minima except in *transition state theory*.

If the equation of $y = x^3$ is plotted, it is possible to see that at $x = 0$ there is a third kind of point, an inflection point, where both $\frac{dy}{dx}$ and $\frac{d^2y}{dx^2}$ are zero.

$$y = f(x) \frac{dy}{dx} = f'(x) \frac{d^2y}{dx^2} = f''(x)$$

Plot $x^3 + x^2 - 6x$ between -4 and +3, in units of 1. (It will speed things up if you factorise it first. Then you will see there are 3 places where $f(x) = 0$ so you only need calculate 5 points.) By factorising you can see that this equation has 3 roots. Find the 2 turning points. (Differentiate once and find the roots of the quadratic equation using $x = -b \dots$. This gives the position of the 2 turning points either side of zero. As the equation is only in x^3 it has 3 roots and 2 maxima / minima at the most therefore we have solved everything.

Differentiate your quadratic again to get $\frac{d^2y}{dx^2}$. Notice that the turning point to the left of zero is a maximum *i.e.* $\frac{d^2y}{dx^2} = -ve$ and the other is a minimum *i.e.* $\frac{d^2y}{dx^2} = +ve$.

What is the solution and the turning point of $y = x^3$.

Solve $x^3 - x = 0$, by factorisation.

(The 3 roots are -3, 0 and +2. $\frac{dy}{dx} = f'(x) = 3x^2 + 2x - 6$

Solutions are $1/3(\sqrt{19} - 1)$ and $-1/3(\sqrt{19} + 1)$, *i.e.* -1.7863 and 1.1196.

$$\frac{d^2y}{dx^2} = f''(x) = 6x + 2$$

There are 3 coincident solutions at $x = 0$, $\frac{d^2y}{dx^2} = 0$, at 0 so this is an *inflection point*.

The roots are 0, 1 and -1.

External links

[1] <http://www.mathtutor.ac.uk/viewdisksinfo.php?disk=6>

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Differentiation&oldid=1359516

Principal Authors: MartinY, Amigadave, Jguk, Whiteknight

Integration

Free Web Based Material from HEFCE

There is a DVD on integration at Math Tutor ^[1].

The basic polynomial

$$f(x) = x^n$$

$$\int f(x)dx = \frac{x^{n+1}}{n+1} + c$$

This works fine for all powers except -1, for instance the integral of

$$\frac{1}{x^7}$$

is just

$$-\frac{1}{6x^6} + c$$

-1 is clearly going to be a special case because it involves an infinity when $x = 0$ and goes to a steep spike as x gets small. As you have learned earlier this integral is the natural logarithm of x and the infinity exists because the *log* of zero is minus infinity and that of negative numbers is undefined.

$$\frac{dx}{dy} = \frac{1}{\frac{dy}{dx}} \quad \text{also} = x$$

therefore

$$\frac{dy}{dx} = \frac{1}{x}$$

$$\int \frac{dy}{dx} dx = y$$

therefore

$$\int \frac{1}{x} dx = \ln x$$

Integrating 1/x like things

Just as

$$\frac{d}{dx} \ln x = \frac{1}{x}$$

so

$$\frac{d}{dx} \ln u = \frac{1}{u} \frac{du}{dx}$$

therefore by the *chain rule*

$$\frac{d}{dx} \ln f(x) = \frac{f'(x)}{f(x)}$$

therefore

$$\int \frac{f'(x)}{f(x)} dx = \ln |f(x)| + c$$

Examples of this are:

$$\int \frac{\cos \phi}{1 + \sin \phi} d\phi = \ln(1 + \sin \phi) + c$$

or

$$\int \frac{\sin \phi}{1 + \cos \phi} d\phi = -\ln(1 + \cos \phi) + c$$

and

$$\int \frac{e^y}{e^y + 2} dy = \ln(e^y + 2) + c$$

As the integral of $1/x$ is \ln so the differential of \ln is $1/x$ so

$$\frac{d}{dx} \ln 5x^3 = \ln 5 + \ln x^3 = \ln 5 + 3 \ln x$$

$\ln 5$ is just a constant so $\frac{d}{dx} = 0$ so

$$\frac{d}{dx} \ln 5x^3 = \frac{3}{x}$$

This can also be done by the chain rule

$$\frac{d}{dx} \ln 5x^3 = \frac{1}{5x^3} \cdot \frac{d(5x^3)}{dx} = \frac{15x^2}{5x^3} = \frac{3}{x}$$

What is interesting here is that the 5 has disappeared completely. The gradient of the log function is unaffected by a multiplier! This is a fundamental property of logs.

Some observations on infinity

Obviously $\frac{1}{0}$ is ∞ .

$\frac{0}{0}$ and $\frac{\infty}{\infty}$ are undefined but sometimes a large number over a large number can have defined values. An example is the sin of 90 degrees, which you will remember has a large opposite over a large hypotenuse but in the limit of an infinitesimally thin triangle they become equal. Therefore the sin is 1.

Definite integrals (limits)

Remember how we do a definite integral

$$\int_0^3 (f(x)) = [F(x)]_0^3 = F(3) - F(0)$$

where F is the indefinite integral of f .

Here is an example where limits are used to calculate the 3 areas cut out by a quartic equation:

$$x^4 - 2x^3 - x^2 + 2x$$

We see that $x = 1$ is a solution so we can do a polynomial division:

```

          x3  -x2  -2x
          -----
x-1 ) x4 -2x3 -x2 +2x
      x4  -x3
      -----
           -x3  -x2
           -x3  +x2
           -----
                -2x2  +2x
                -2x2  +2x
                -----
                       0
    
```

So the equation is $x(x^2 - x - 2)(x - 1)$ which factorises to $x(x - 2)(x + 1)(x - 1)$.

$$\int_a^b x^4 - 2x^3 - x^2 + 2x = \left[\frac{x^5}{5} - \frac{x^4}{2} - \frac{x^3}{3} + x^2 \right]_a^b$$

Integration by substitution

$$\int \sin \theta \cos \theta d\theta = \int u \frac{du}{d\theta} d\theta$$

where $u = \sin \theta$.

$$\int \sin \theta \cos \theta d\theta = \frac{\sin^2 \theta}{2} + c$$

$$\int \sin^9 \theta \cos \theta d\theta \quad \text{similarly} \quad = \frac{\sin^{10} \theta}{10} + c$$

Simple integration of trigonometric and exponential Functions

$$\int 4e^{2x} dx$$

$$\int -9e^{-3x} dx$$

$$\int \sin 2\theta d\theta$$

$$\int \sin^2 \theta + \cos^2 \theta d\theta$$

$$\int -\cos \phi d\phi$$

Answers

$$2e^{2x} + c$$

$$3e^{-3x} + c$$

$$-\frac{1}{2} \cos 2\theta + c$$

$$\theta + c$$

i.e. the integral of $1d\theta$.

$$\sin \phi + c$$

Integration by parts

This is done in many textbooks and wikipedia. Their notation might be different to the one used here, which hopefully is the most clear. You derive the expression by taking the product rule and integrating it. You then make one of the UV' into a product itself to produce the expression.

$$\int UV = U \int V - \int (U' \int V)$$

(all integration with respect to x . Remember by

$$\int UV = U[\text{int}] - \int [\text{diff}][\text{int}]$$

(U gets differentiated.)

The important thing is that you have to integrate one expression of the product and differentiate the other. In the basic scheme you integrate the most complicated expression and differentiate the simplest. Each time you do this you generate a new *term* but the

function being differentiated goes to zero and the integral is solved. The expression which goes to zero is U .

The other common scheme is where the parts formula generates the expression you want on the right of the equals and there are no other integral signs. Then you can rearrange the equation and the integral is solved. This is obviously very useful for trig functions where $\sin \gg \cos \gg -\sin \gg -\cos \gg \sin \text{ ad infinitum}$.

e^x also generates itself and is susceptible to the same treatment.

$$\begin{aligned} \int e^{-x} \sin x \, dx &= (-e^{-x}) \sin x - \int (-e^{-x}) \cos x \, dx \\ &= -e^{-x} \sin x + \int e^{-x} \cos x \, dx \\ &= -e^{-x}(\sin x + \cos x) - \int e^{-x} \sin x \, dx + c \end{aligned}$$

We now have our required integral on both sides of the equation so

$$= -\frac{1}{2}e^{-x}(\sin x + \cos x) + c$$

Integration Problems

Integrate the following by parts with respect to x .

$$xe^x$$

$$x^2 \sin x$$

$$x \cos x$$

$$x^2 e^x$$

$$x^2 \ln x$$

$$e^x \sin x$$

$$e^{2x} \cos x$$

$$x(1+x)^7$$

Actually this one can be done quite elegantly by parts, to give a two term expression. Work this one out. Expanding the original integrand by Pascal's Triangle gives:

$$\begin{array}{cccccccc} & & 2 & & 3 & & 4 & & 5 & & 6 & & 7 & & 8 \\ & & x & + & 7x & + & 21x^2 & + & 35x^3 & + & 35x^4 & + & 21x^5 & + & 7x^6 & + & x^7 \end{array}$$

The two term integral expands to

$$\begin{array}{ccccccccccc} & & 2 & & 3 & & 4 & & 5 & & 6 & & 7 & & 8 & & 9 \\ & & 1/2 x & + & 7/3 x^2 & + & 21/4 x^3 & + & 7x^4 & + & 35/6 x^5 & + & 3x^6 & + & 7/8 x^7 & + & 1/9 x^8 & - & 1/72 x^9 \end{array}$$

So one can see it is correct on a term by term basis.

$$3x \sin x$$

$$2x^2 \cos 2x$$

If you integrate $x^7 \sin x$ you will have to apply *parts* 7 times to get x to become 1 thereby generating 8 terms:

$$\begin{array}{ccccccccc} & & 7 & & 6 & & 5 & & 4 & & 3 \\ & & -x \cos(x) & + & 7x \sin(x) & + & 42x^2 \cos(x) & - & 210x^3 \sin(x) & - & 840x^4 \cos(x) & + \end{array}$$

$$2520 x^2 \sin(x) + 5040 x \cos(x) - 5040 \sin(x) + c$$

(Output from Maple.)

Though it looks nasty there is quite a pattern to this, $7, 7 \times 6, 7 \times 6 \times 5 \dots 7!$ and $\sin, \cos, -\sin, -\cos, \sin, \cos$ etc so it can easily be done by hand.

Differential equations

First order differential equations are covered in many textbooks. They are solved by integration. (First order equations have $\frac{dy}{dx}$, second order equations have $\frac{dy}{dx}$ and $\frac{d^2y}{dx^2}$.)

The arbitrary constant means another piece of information is needed for complete solution, as with the Newton's Law of Cooling and Half Life examples.

Provided all the x s can be got to one side and the y s to another the equation is separable.

$$2y \frac{dy}{dx} = 6x^2$$

$$\int 2y dy = \int 6x^2 dx$$

$$y^2 = 2x^3 + c$$

This is the *general solution*.

Typical examples are:

$$y = x \frac{dy}{dx} \quad \int \frac{dy}{y} = \int \frac{dx}{x}$$

$$\ln y = \ln x + \ln A \quad (\text{constant}) \quad \text{i.e.} \quad y = Ax$$

by definition of logs.

$$\frac{dy}{dx} = ky \quad (1)$$

$$kxy = \frac{dy}{dx}$$

$$\int \frac{dy}{y} = \int kx dx \quad (2)$$

$$\ln y = \frac{1}{2} kx^2 + A \quad (3)$$

This corresponds to:

$$y = B e^{\frac{1}{2} kx^2}$$

The Schrödinger equation is a 2nd order differential equation *e.g.* for the particle in a box

$$\frac{\hbar^2}{2m} \frac{d^2\Psi}{dx^2} + E\Psi = 0$$

It has taken many decades of work to produce computationally efficient solutions of this equation for polyatomic molecules. Essentially one expands in coefficients of the *atomic orbitals*. Then integrates to make a differential equation a set of numbers, integrals, in a

matrix. Matrix algebra then finishes the job and finds a solution by solving the resultant simultaneous equations.

The calculus of trigonometric functions

There are many different ways of expressing the same thing in trig functions and very often successful integration depends on recognising a trig identity.

$$\int \sin 2x dx = -\frac{1}{2} \cos 2x + c$$

but could also be

$$\sin^2 x \text{ or } -\cos^2 x$$

(each with an integration constant!).

When applying calculus to these functions it is necessary to spot which is the simplest form for the current manipulation. For integration it often contains a product of a function with its derivative like $2 \sin x \cos x$ where integration by substitution is possible.

Where a derivative can be spotted on the numerator and its integral below we will get a \ln function. This is how we integrate \tan .

$$\int \tan x dx = \int \frac{\sin x}{\cos x} dx = -\ln (\cos x) + c$$

We can see this function goes to infinity at $\pi/2$ as it should do.

Integration by rearrangement

Take for example:

$$\int \cos^2 x dx$$

Here there is no \sin function produced in with the \cos powers so we cannot use substitution. However there are the two trig identities

$$\cos^2 \theta + \sin^2 \theta = 1$$

and

$$\cos^2 \theta - \sin^2 \theta = \cos 2\theta$$

Using these we have

$$\int \cos^2 x dx = \int \frac{1}{2}(1 + \cos 2x) dx$$

so we have two simple terms which we can integrate.

The Maclaurin series

We begin by making the assumption that a function can be approximated by an infinite power series in x :

$$f(x) = a_0 + a_1 x + a_2 x^2 + a_3 x^3 + a_4 x^4 + \dots$$

By differentiating and setting $x = 0$ one gets

$$f(x) = f(0) + f'(0)x + \frac{f''(0)}{2!}x^2 + \frac{f'''(0)}{3!}x^3 + \frac{f^{(4)}(0)}{4!}x^4 + \dots$$

\sin , \cos and e^x can be expressed by this series approximation

$$\sin x = x - \frac{x^3}{3!} + \frac{x^5}{5!} - \frac{x^7}{7!} \dots$$

$$\cos x = 1 - \frac{x^2}{2!} + \frac{x^4}{4!} - \frac{x^6}{6!} \dots$$

$$e^x = 1 + x + \frac{x^2}{2!} + \frac{x^3}{3!} + \frac{x^4}{4!} \dots$$

Notice e^x also works for negative x .

When differentiated or integrated e^x generates itself!

When differentiated $\sin x$ generates $\cos x$.

By using series we can convert a complex function into a polynomial, and can use $\sin x = x - x^3/6$ for small x .

In actual fact the kind of approximation used inside computer programs is more like:

$$\frac{ax^2 + bx + c}{Ax^2 + Bx + C}$$

These have greater range but are much harder to develop and a bit fiddly on the calculator or to estimate by raw brain power.

We cannot expand $\ln x$ this way because $\ln 0$ is

$-\infty$. However $\ln(1+x)$ can be expanded.

Work out the series for $\ln(1+x)$.

Factorials

The factorials you have seen in series come from repeated differentiation. $n!$ also has a statistical meaning as it is the number of unique ways you can arrange n objects.

$0!$ is 1 by definition, *i.e.* the number of different ways you can arrange 0 objects is 1.

In statistical thermodynamics you will come across many factorials in expressions such as:

$$W = \frac{N!}{n_0!n_1!n_2!\dots}$$

Factorials rapidly get unreasonably large: $6! = 720$, $8! = 40320$ but $12! = 479001600$ so we need to divide them out into reasonable numbers if possible, so for example $8!/6! = 7 \times 8$.

Stirling's approximation

Also in statistical thermodynamics you will find *Stirling's approximation*:

$$\ln x! \approx x \ln x - x$$

This is proved and discussed in Atkins' *Physical Chemistry*.

How can you use series to estimate $\ln 2$. Notice that the series for

$\ln(1+x)$ converges extremely slowly. e^x is much faster because the $n!$ denominator becomes large quickly.

Trigonometric power series

Remember that when you use $\sin x = x$ and $\cos x = 1 - x^2/2$ that x must be in radians.....

Calculus revision

Problems

1. Differentiate $e^{-t} \cos 2t$, with respect to t . (Hint - use the chain rule.)
2. Differentiate $x^2(3x + 1)^4$. (Chain rule and product rule here.)
3. Differentiate $\ln(7x^4)$. (Hint - split it into a sum of logs first.)
4. Integrate $\ln x$. (Hint - use integration by parts and take the expression to be differentiated as 1.)

Answers

1. It is just $e^{-t} \cdot -2 \sin 2t - e^{-t} \cos 2t$. Bring a $-e^{-t}$ out of each term to simplify to $-e^{-t}(2 \sin 2t + \cos 2t)$.
2. $2x(9x + 1)(3x + 1)^3$.
3. $\ln 7 + 4 \ln x$ - therefore it is 4 times the derivative of $\ln x$.
4. You should get $x \ln x - x$ by 1 application of *parts*.

External links

[1] <http://www.mathtutor.ac.uk/viewdisksinfo.php?disk=7>

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Integration&oldid=1359382

Principal Authors: MartinY, Amigadave, Jake Wasdin, Jguk, Whiteknight

Some useful aspects of calculus

Limits

Many textbooks go through the proper theory of differentiation and integration by using limits. As chemists it is possible to live without knowing this so we might well not have it as an examinable topic. However here is how we differentiate **sin** from 1st principles.

$$\begin{aligned} \frac{dy}{dx} &= \left(\frac{\delta y}{\delta x} \right)_{\text{limit } \delta x \rightarrow 0} \\ &= \frac{\sin(x + \delta x) - \sin x}{\delta x} \\ &= \frac{\sin x \cos \delta x + \sin \delta x \cos x - \sin x}{\delta x} \\ &= \frac{\sin x}{\delta x} - \frac{\sin x}{\delta x} + \frac{\sin \delta x}{\delta x} \cos x \end{aligned}$$

As $\sin \delta x = \delta x$ for small x this expression is $\cos x$.

Similarly for $\cos \theta$

$$\begin{aligned}\frac{dy}{d\theta} &= \frac{\cos(\theta + \delta\theta) - \cos\theta}{\delta\theta} \\ &= \frac{\cos\theta \cos\delta\theta - \sin\theta \sin\delta\theta - \cos\theta}{\delta\theta} \\ &= \frac{\cos\theta}{\delta\theta} - \frac{\sin\theta \delta\theta}{\delta\theta} - \frac{\cos\theta}{\delta\theta}\end{aligned}$$

This is equal to $-\sin\theta$.

Numerical differentiation

You may be aware that you can fit a quadratic to 3 points, a cubic to 4 points, a quartic to 5 *etc.* If you differentiate a function numerically by having two values of the function δx apart you get an approximation to $\frac{dy}{dx}$ by constructing a triangle and the gradient is the tangent. There is a forward triangle and a backward triangle depending on the sign of δx . These are the forward and backward differentiation approximations.

If however you have a central value with a δ either side you get the central difference formula which is equivalent to fitting a quadratic, and so is second order in the small value of δx giving high accuracy compared with drawing a tangent. It is possible to derive this formula by fitting a quadratic and differentiating it to give:

$$\frac{dy}{dx} = \frac{f^+ + f^- - 2f^0}{\delta^2}$$

HCl	r-0.02			
sigma (iso)		32.606716	142.905788	-110.299071
HCl	r-0.01			
sigma (iso)		32.427188	142.364814	-109.937626
HCl	r0	Total shielding: paramagnetic : diamagnetic		
sigma (iso)		32.249753	141.827855	-109.578102
HCl	r+0.01			
sigma (iso)		32.074384	141.294870	-109.220487
HCl	r+0.02			
sigma (iso)		31.901050	140.765819	-108.864769

This is calculated data for the shielding in ppm of the proton in HCl when the bondlength is stretched or compressed by 0.01 of an Angstrom (not the approved unit pm). The total shielding is the sum of two parts, the paramagnetic and the diamagnetic. Notice we have retained a lot of significant figures in this data, always necessary when doing numerical differentiation.

Exercise - use numerical differentiation to calculate $d(\text{sigma}) / dr$ and $d^2(\text{sigma}) / dr^2$ using a step of 0.01 and also with 0.02. Use 0.01 to calculate $d(\text{sigma(para)}) / dr$ and $d(\text{sigma(dia)}) / dr$.

Numerical integration

Wikipedia has explanations of the Trapezium rule and Simpson's Rule. Later you will use computer programs which have more sophisticated versions of these rules called Gaussian quadratures inside them. You will only need to know about these if you do a numerical project later in the course. Chebyshev quadratures are another version of this procedure which are specially optimised for integrating noisy data coming from an experimental source. The mathematical derivation averages rather than amplifies the noise in a clever way.

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Some_Useful_Aspects_of_Calculus&oldid=1359399

Principal Authors: MartinY, Amigadave, Jguk, Whiteknight

Tests and examples

Some mathematical examples applied to chemistry

Variable names

The ubiquitous x is not always the variable as you will all know by now. One problem dealing with real applications is sorting out which symbols are the variables and which are constants. (If you look very carefully at professionally set equations in text books you should find that there are rules that constants are set in Roman type, *i.e.* straight letters and variables in *italics*. Do not rely on this as it is often ignored.)

Here are some examples where the variable is conventionally something other than x .

1. The Euler angles which are used in rotation are conventionally α, β and γ not the more usual angle names θ and ϕ . The *rotation matrix* for the final twist in the commonest

Euler definition therefore looks like
$$\begin{bmatrix} \cos(\gamma) & \sin(\gamma) & 0 \\ -\sin(\gamma) & \cos(\gamma) & 0 \\ 0 & 0 & 1 \end{bmatrix}$$

2. The energy transitions in the hydrogen atom which give the Balmer series are given by the formula $\tilde{\nu} = R_H \left(\frac{1}{2^2} - \frac{1}{n^2} \right)$ $\tilde{\nu}$ is just a single variable for the energy the *tilde* being a convention used by spectroscopists to say it is wavenumbers, (cm^{-1}). The H subscript on R_H has no mathematical meaning. It is the Rydberg constant and is therefore in Roman type. R_H is known very accurately as $109,677.581 \text{ cm}^{-1}$. It has actually been known for a substantial fraction of the class to make an error putting this fraction over a common denominator in examination conditions.

3. In the theory of light ω is used for frequency and t not surprisingly for time. Light is an oscillating electric and magnetic field therefore the cosine function is a very good way of describing it. You can also see here the use of complex numbers. Using the real axis of the Argand diagram^[1] for the electric field and the imaginary axis for the magnetic field is a very natural description mathematically and maps ideally onto the physical reality. Here we are integrating with respect to t and ω , the operating frequency if it is a laser experiment is a constant, therefore it appears on the denominator in the integration.

$$\int \cos(\omega t) \sin(\omega t) dt = -\frac{\cos(\omega)^2}{2\omega} + c$$

In this case we can see a physical interpretation for the integration constant. It will be a *phase factor*. If we were dealing with sunlight we might well be integrating a different function over ω in order to calculate all of the phenomenon which has different strengths at the different light frequencies. Our integration limits would either be from zero to infinity or perhaps over the range of energies which correspond to visible light.

4. This example is a laser experiment called Second Harmonic Generation. There is an electric field F , frequency ν and a property constant $\chi^{(2)}$. ϵ_0 is a fundamental constant. We have an intense monochromatic laser field fluctuating at the frequency ν , (*i.e.* a
-

strong light beam from a big laser). $F_v \sin(2\pi\nu t)$ Therefore the F^2 term contributes $\epsilon_0 \chi^{(2)} F_v^2 \sin^2(2\pi\nu t)$ to the polarization. We know from trigonometric identities that \sin^2 can be represented as a cosine of the double angle $\cos(2\theta) = 1 - 2\sin^2\theta$ Therefore the polarization is $\frac{1}{2}\epsilon_0 \chi^{(2)} F_v^2 (1 - \cos 4\pi\nu t)$ In this forest of subscripts and Greek letters the important point is that there are *two* terms contributing to the output coming from $(1 - \cos 4\pi\nu t)$ which multiplies the rest of the stuff. In summary we have $A \sin^2(t)$ is equal to $B(1 - \cos(2t))$ where everything except the $\text{trig}(t)$ and $\text{trig}(2t)$ are to some extent unimportant for the phenomenon of doubling the frequency. \sin and \cos differ only in a phase shift so they represent the same physical phenomenon, *i.e.* light, which has phase. (One of the important properties of laser light is that it is *coherent*, *i.e.* it all has the same phase. This is fundamentally embedded in our mathematics.)

van der Waals Energy

The van der Waals energy between two inert gas atoms can be written simply as a function of r

$$E_{vdw} = \frac{A}{r^{12}} - \frac{B}{r^6}$$

Notice that the r^{12} term is positive corresponding to repulsion. The r^6 term is the attractive term and is negative corresponding to a reduction in energy. A and B are constants fitted to experimental numbers.

This function is very easy to both differentiate and integrate. Work these out. In a gas simulation you would use the derivative to calculate the forces on the atoms and would integrate Newton's equations to find out where the atoms will be next.

Another potential which is used is:

$$E_{vdw} = Ae^{-Br} - Cr^{-6}$$

This has 1 more fittable constant. Integrate and differentiate this.

The $\frac{A}{r^{12}} - \frac{B}{r^6}$ is called a Lennard-Jones potential and is often expressed using the 2 parameters of an energy ϵ and a distance σ .

$$E(r) = 4\epsilon \left(\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right)$$

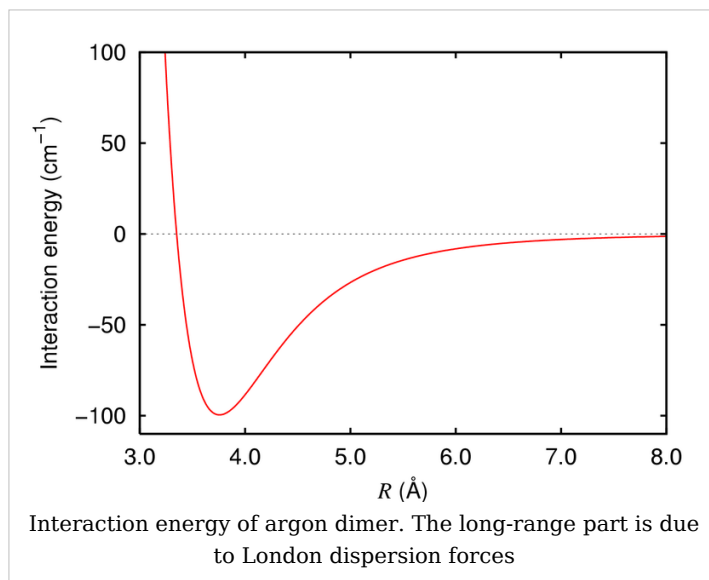
ϵ is an energy. Set the derivative of this to zero and find out where the van der Waals minimum is. Differentiate again and show that the derivative is positive, therefore the well is a minimum, not a turning point.

A diatomic potential energy surface

In a diatomic molecule the energy is expanded as the bond stretches in a polynomial. We set $x = (r - r_0)$. At r_0 the function is a minimum so there is no dE/dx term.

$$E = \frac{1}{2}k_{\text{harm.}} \frac{d^2 E}{dx^2} + \frac{1}{6}k_{\text{anharm.}} \frac{d^3 E}{dx^3}$$

Whatever function is chosen to provide the energy setting the 1st derivative to zero will be required to calculate r_0 . The 2nd and 3rd derivatives will then need to be evaluated to give the shape of the potential and hence the infra-red spectrum. E is usually modelled by a very complicated function whose differentiation is not entered into lightly.



A one-dimensional metal

A one-dimensional metal is modelled by an infinite chain of atoms 150 picometres apart. If the metal is lithium each nucleus has charge 3 and its electrons are modelled by the function

$$\cos^2 \left\{ \frac{\pi r}{2 \cdot 150} \right\}$$

which repeats every 150 pm. What constant must this function be multiplied by to ensure there are 3 electrons on each atom? (Hint... integrate \cos^2 between either $-\frac{\pi}{2}$ and $+\frac{\pi}{2}$ or -75pm and +75pm according to your equation. This integral is a dimensionless number equal to the number of electrons, so we will have to multiply by a *normalisation constant*.)

Here we have modelled the density of electrons. Later in the second year you will see electronic structure more accurately described by functions for *each independent electron* called *orbitals*. These are subject to rigorous mathematical requirements which means they are quite fun to calculate.

Kepler's Laws

Another physics problem but a good example of a log-log plot is the radius and time period relations of the planets.

This data is dimensionless because we have divided by the time / distance of the earth. We can take logs of both.

	Mercury	Venus	Earth	Mars	Jupiter	Saturn
r	0.3871	0.7233	1	1.524	5.203	9.539
T	0.2408	0.6152	1	1.881	11.86	29.46

	Mercury	Venus	Earth	Mars	Jupiter	Saturn
$\log_{10} r$	-0.4122		0			0.9795
$\log_{10} T$	-0.6184		0			1.4692

Try a least squares fit on your spreadsheet program. Using the Earth and Saturn data: (which is extremely bad laboratory practice, to use just two points from a data set!)

$$\Delta(\log T) / \Delta(\log r) = 1.5000359$$

$$\text{so } \log T = 1.5 \log r = \log r^{1.5}$$

$$\text{so } T = r^{3/2} \text{ and } T^2 = r^3$$

This is Kepler's 3rd law. If you use either a least squares fit gradient or the mercury to saturn data you get the same powers. We have got away with not using a full data set because the numbers given are unusually accurate and to some extent tautological, (remember the planets go round in ellipses not circles!).

Newton's law of cooling

θ is the excess temperature of a cooling body over room temperature (20°C say). The rate of cooling is proportional to the excess temperature.

$$-\frac{d\theta}{dt} = k\theta$$

$$\text{using } \frac{dx}{dy} = \frac{1}{\frac{dy}{dx}}$$

$$\frac{dt}{d\theta} = -\frac{1}{k\theta}$$

This is a differential equation which we integrate with respect to θ to get

$$t = -\frac{1}{k} \ln \theta + c$$

The water is heated to 80° C and room temperature is 20° C. At the beginning $t = 0$ and $\theta = 60$, so

$$0 = -\frac{1}{k} \ln 60 + c$$

therefore

$$c = \frac{\ln 60}{k}$$

$$t = -\frac{1}{k} \ln \theta + \frac{1}{k} \ln 60$$

$$\text{but } \ln 60 - \ln \theta = -\ln \frac{\theta}{60}$$

$$\text{so } \ln \frac{\theta}{60} = -kt$$

After 5 minutes the water has cooled to 70° C.

$$\text{so } \ln \frac{50}{60} = -5k \text{ so } k = -\frac{1}{5} \ln \frac{5}{6} = \frac{1}{5} \ln \frac{6}{5} . \text{ So } k = 0.0365$$

$$\ln \frac{\theta}{60} = -0.0365t$$

$$\frac{\theta}{60} = e^{-0.0365t}$$

by the definition of logarithms. This gives the plot of an exponential decay between 80 and 20° C.

So after 10 minutes $t = 20 + 60e^{-0.365} = 61.6^\circ$ C. After 20 minutes

$$t = 20 + 60e^{-0.73} = 49.9^\circ \text{ C. After 30 minutes } t = 20 + 60e^{-1.10} = 40.1^\circ \text{ C.}$$

Bacterial Growth

2 grams of an organism grows by $1/10$ gram per day per gram.

$$\frac{dm}{dt} = \frac{m}{10}$$

This is a differential equation which is solved by integration thus

$$\int \frac{10}{m} dm = \int dt$$

$$10 \ln m = t + c$$

$$\ln m = \frac{t}{10} + c$$

therefore

$$m = e^c e^{\frac{t}{10}}$$

When $t = 0$ we have 2 grams so

$$m = 2e^{t/10}$$

For the sample to double in mass

$$4 = 2e^{t/10}$$

$$2 = e^{t/10}$$

$$\ln 2 = \frac{t}{10}$$

$$t = 10 \ln 2 = 6.9315 \text{ days}$$

Half life calculations are similar but the exponent is negative.

Partial fractions for the 2nd order rate equation

In chemistry work you will probably be doing the 2nd order rate equation which requires partial fractions in order to do the integrals.

If you remember we have something like

$$\frac{1}{(2-x)(3-x)} = \frac{A}{(2-x)} + \frac{B}{(3-x)}$$

Put the right-hand side over a common denominator

$$\frac{1}{(2-x)(3-x)} = \frac{A(3-x) + B(2-x)}{(2-x)(3-x)}$$

This gives $1 = 3A - Ax + 2B - Bx$

By setting x to 3 we get $1 = -B$ ($B=-1$). Setting $x = 0$ and $B = -1$

$$1 = 3A - 2 \quad (A=+1)$$

$$\text{Check} \quad 1 = 3 - x - 2 + x \quad \text{true...}$$

Therefore

$$\int \frac{1}{(2-x)(3-x)} dx = \int \frac{1}{(2-x)} - \int \frac{1}{(3-x)} dx = \ln(2-x) + \ln(3-x) + c$$

noting the sign changes on integrating $1/(2-x)$ not $1/x$.

External links

[1] <http://mathworld.wolfram.com/ArgandDiagram.html>

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Some_Mathematical_Examples_applied_to_Chemistry&oldid=1359395

Principal Authors: MartinY, Amigadave, Jguk, Whiteknight

Tests and exams

A possible final test with explanatory notes

This test was once used to monitor the broad learning of university chemists at the end of the 1st year and is intended to check, somewhat lightly, a range of skills in only 50 minutes. It contains a mixture of what are perceived to be both easy and difficult questions so as to give the marker a good idea of the student's algebra skills and even whether they can do the infamous *integration by parts*.

(1) Solve the following equation for x

$$x^2 + 2x - 15 = 0$$

It factorises with 3 and 5 so : $(x + 5)(x - 3) = 0$ therefore the **roots** are -5 and +3, not 5 and -3!

(2) Solve the following equation for x

$$2x^2 - 6x - 20 = 0$$

Divide by 2 and get $x^2 - 3x - 10 = 0$.

This factorises with 2 and 5 so : $(x - 5)(x + 2) = 0$ therefore the **roots** are 5 and -2.

(3) Simplify

$$\ln w^6 - 4 \ln w$$

Firstly $6 \ln w - 4 \ln w$ so it becomes $2 \ln w$.

(4) What is

$$\log_2 \frac{1}{64}$$

$64 = 8 \times 8$ so it also equals $2^3 \times 2^3$ i.e. $\frac{1}{64}$ is 2^{-6} , therefore the answer is -6.

(5) Multiply the two complex numbers

$$3 + 5i \quad \text{and} \quad 3 - 5i$$

These are complex conjugates so they are 3^2 minus $i^2 \times 5^2$ i.e. plus 25 so the total is 34.

(6) Multiply the two complex numbers

$$(5, -2) \quad \text{and} \quad (-5, -2)$$

The real part is -25 plus the $4i^2$. The cross terms make $-10i$ and $+10i$ so the imaginary part disappears.

(7) Differentiate with respect to x :

$$\frac{1}{3x^2} - 3x^2$$

$$\text{Answer: } -\frac{2}{3x^3} - 6x$$

$$(8) \frac{6}{x^4} + 3x^3$$

$$\text{Answer: } 9x^2 - \frac{24}{x^5}$$

$$(9) \frac{2}{\sqrt{x}} + 2\sqrt{x}$$

$$\text{Answer: } \frac{1}{\sqrt{x}} - \frac{1}{\sqrt{x^3}}$$

$$(10) x^3(x - (2x + 3)(2x - 3))$$

Expand out the difference of 2 squares first.....collect and multiply....then just differentiate term by term giving: $20x^4 - 4x^3 + 27x^2$

(11) $3x^3 \cos 3x$

This needs the product rule.... Factor out the $9x^2$ $9x^2(\cos 3x - x \sin 3x)$

(12) $\ln(1-x)^2$

This could be a chain rule problem..... $\frac{1}{(1-x)^2} \cdot 2 \cdot (-1) \cdot (1-x)$

or you could take the power 2 out of the log and go straight to the same answer with a shorter version of the chain rule to: $-\frac{2}{(1-x)}$.

(13) Perform the following integrations:

$$\int (2 \cos^2 \theta + 2\theta) d\theta$$

\cos^2 must be converted to a double angle form as shown many times.... then all 3 bits are integrated giving

$$\cos \theta \sin \theta + \theta + \theta^2$$

(14) $\int \left(8x^{-3} - \frac{4}{x} + \frac{8}{x^3} \right) dx$

Apart from $-\frac{4}{x}$, which goes to \ln , this is straightforward polynomial integration. Also

there is a nasty trap in that two terms can be telescoped to $\frac{16}{x^3}$.

$$-\left(\frac{8}{x^2} + 4 \ln x \right)$$

(15) What is the equation corresponding to the determinant:

$$\begin{vmatrix} b & \frac{1}{\sqrt{2}} & 0 \\ \frac{1}{\sqrt{2}} & b & 1 \\ 0 & 1 & b \end{vmatrix} = 0$$

The first term is $b(b^2 - 1)$ the second $-\frac{1}{\sqrt{2}}\left(\frac{b}{\sqrt{2}} - 0\right)$ and the 3rd term zero. This adds up to $b^3 - 3/2b$.

(16) What is the general solution of the following differential equation:

$$\frac{d\phi}{dr} = \frac{A}{r}$$

where A is a constant..

$$\theta = A \ln r + k .$$

(17) Integrate by parts: $\int x \sin x dx$

Make x the factor to be differentiated and apply the formula, taking care with the signs...

$$\sin x - x \cos x .$$

(18) The Maclaurin series for which function begins with these terms?

$$1 + x + x^2/2! + x^3/3! + x^4/4! + \dots$$

It is e^x

(19) Express

$$\frac{x-2}{(x-3)(x+4)}$$
 as partial fractions.

It is $\frac{1}{7(x-3)} + \frac{6}{7(x+4)}$

(20) What is $2e^{i4\phi} - \cos 4\phi$ in terms of \sin and \cos

This is just Euler's equation..... $2e^{i4\phi} = 2\cos 4\phi - 2i\sin 4\phi$

so one $\cos 4\phi$ disappears to give ... $\cos 4\phi - 2i\sin 4\phi$.

50 Minute Test II

(1) Simplify $2\ln(1/x^3) + 5\ln x$

(2) What is $\log_{10} \frac{1}{10\,000}$

(3) Solve the following equation for t

$$t^2 - 3t - 4 = 0$$

(4) Solve the following equation for w

$$w^2 + 4w - 12 = 0$$

(5) Multiply the two complex numbers $(-4, 3)$ and $(-5, 2)$

(6) Multiply the two complex numbers $3 + 2i$ and $3 - 2i$

(7) The Maclaurin series for which function begins with these terms?

$$x - x^3/6 + x^5/120 + \dots$$

(8) Differentiate with respect to x :

$$x^3(2 - 3x)^2$$

(9) $\frac{\sqrt{x}}{2} - \frac{\sqrt{3}}{2\sqrt{x}}$

(10) $x^4 - 3x^2 + k$

where k is a constant.

(11) $\frac{2}{3x^4} - Ax^4$

where A is a constant.

(12) $3x^3 e^{3x}$

(13) $\ln(2 - x)^3$

(14) Perform the following integrations:

$$\int \left(3w^4 - 2w^2 + \frac{6}{5w^2} \right) dw$$

(15) $\int (3 \cos \theta + \theta) d\theta$

(16) What is the equation belonging to the determinant $\begin{vmatrix} x & 0 & 0 \\ 0 & x & i \\ 0 & i & x \end{vmatrix} = 0$

(17) What is the general solution of the following differential equation:

$$\frac{dy}{dx} = ky$$

(18) Integrate by any appropriate method:

$$\int \left(\ln x + \frac{4}{x} \right) dx$$

(19) Express $\frac{x+1}{(x-2)(x+2)}$ as partial fractions.

(20) What is $2e^{i2\phi} + 2i \sin 2\phi$ in terms of sin and cos.

50 Minute Test III

(1) Solve the following equation for t

$$t^2 - 4t - 12 = 0$$

(2) What is $\log_4 \frac{1}{16}$

(3) The Maclaurin series for which function begins with these terms?

$1 - x^2/2 + x^4/24 + \dots$ ---- (4) Differentiate with respect to x :

$$\frac{5}{x^2} - 8x^4$$

(5) $\frac{4}{\sqrt{x}} - \sqrt{2}x$

(6) $5\sqrt{x} + \frac{6}{x^3}$

(7) $\frac{5}{x^3} - 5x^3$

(8) $x^2(2x^2 - (5 + 2x)(5 - 2x))$

(9) $2x^2 \sin x$

(10) Multiply the two complex numbers $(2, 3)$ and $(2, -3)$

(11) Multiply the two complex numbers $3 - i$ and $-3 + i$

(12) Perform the following integrations:

$$\int \left(\frac{1}{3x} + \frac{1}{3x^2} - 5x^{-6} \right) dx$$

(13)

$$\int \left(6x^{-2} + \frac{2}{x} - \frac{8}{x^2} \right) dx$$

(14) $\int (\cos^2 \theta + \theta) d\theta$

(15) $\int (\sin^3 \theta \cos \theta + 2\theta) d\theta$

(16) Integrate by parts: $\int 2x \cos x dx$

(17) What is the equation corresponding to the determinant:

$$\begin{vmatrix} x & -1 & 0 \\ -1 & x & 0 \\ 0 & 0 & x \end{vmatrix} = 0$$

(18) Express $\frac{x-1}{(x+3)(x-4)}$ as partial fractions.

(19) What is the general solution of the following differential equation:

$$\frac{d\theta}{dr} = \frac{r}{A}$$

(20) What is $e^{i2\phi} - 2i \sin 2\phi$ in terms of sin and cos.

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Tests_and_Exams&oldid=1359441

Principal Authors: MartinY, Amigadave, Whiteknight

Further reading

Further reading

Further reading

Books

- Paul Monk, *Maths for Chemistry: A Chemist's Toolkit of Calculations*, (Oxford, 2006), ISBN 978-0199277414.
- Bostock and Chandler, *Core Maths for A-level, Third Edition*, (Nelson Thornes, 2000), ISBN 978-0748755097.
- M. C. R. Cockett and G. Doggett, *Maths for Chemists: Numbers, Functions and Calculus v. 1*, (Royal Society of Chemistry, London, 2003), ISBN 978-0854046775
- M. C. R. Cockett and G. Doggett, *Maths for Chemists: Power Series Complex Numbers and Linear Algebra v. 2*, (Royal Society of Chemistry, London, 2003), ISBN 978-0854044955
- G. Currell and T. Dowman, *Mathematics and Statistics for Science*, (Wiley, 2005), ISBN 978-0470022290.
- Stephen K. Scott, *Beginning Maths for Chemistry*, (Oxford, 1995), ISBN 978-0198559306
- Peter Tebbutt, *Basic Mathematics for Chemists, 2nd edition*, (Wiley, 1994), ISBN 978-0471972839

Online resources

There is much useful free material relevant to this book, including downloadable DVDs, funded by the HEFCE Fund for the Development of Teaching & Learning and the Gatsby Technical Education Project in association with the Higher Education Academy at Math Tutor ^[1].

External links

[1] <http://www.mathtutor.ac.uk/>

Source: http://en.wikibooks.org/w/index.php?title=Mathematics_for_chemistry/Further_reading&oldid=1359358

Principal Authors: Amigadave

License

GNU Free Documentation License

1. REDIRECT Wikibooks:GNU Free Documentation License

Source: http://en.wikibooks.org/w/index.php?title=GNU_Free_Documentation_License&oldid=320579

Principal Authors: Guanaco

License

Shortcut:WP:GFDL

Version 1.2, November 2002

Copyright (C) 2000,2001,2002 Free Software Foundation, Inc. 51 Franklin St, Fifth Floor, Boston, MA 02110-1301 USA Everyone is permitted to copy and distribute verbatim copies of this license document, but changing it is not allowed.

0.PREAMBLE

The purpose of this License is to make a manual, textbook, or other functional and useful document "free" in the sense of freedom: to assure everyone the effective freedom to copy and redistribute it, with or without modifying it, either commercially or noncommercially. Secondly, this License preserves for the author and publisher a way to get credit for their work, while not being considered responsible for modifications made by others.

This License is a kind of "copyleft", which means that derivative works of the document must themselves be free in the same sense. It complements the GNU General Public License, which is a copyleft license designed for free software.

We have designed this License in order to use it for manuals for free software, because free software needs free documentation: a free program should come with manuals providing the same freedoms that the software does. But this License is not limited to software manuals; it can be used for any textual work, regardless of subject matter or whether it is published as a printed book. We recommend this License principally for works whose purpose is instruction or reference.

1.APPLICABILITY AND DEFINITIONS

This License applies to any manual or other work, in any medium, that contains a notice placed by the copyright holder saying it can be distributed under the terms of this License. Such a notice grants a world-wide, royalty-free license, unlimited in duration, to use that work under the conditions stated herein. The "Document", below, refers to any such manual or work. Any member of the public is a licensee, and is addressed as "you". You accept the license if you copy, modify or distribute the work in a way requiring permission under copyright law.

A "Modified Version" of the Document means any work containing the Document or a portion of it, either copied verbatim, or with modifications and/or translated into another language.

A "Secondary Section" is a named appendix or a front-matter section of the Document that deals exclusively with the relationship of the publishers or authors of the Document to the Document's overall subject (or to related matters) and contains nothing that could fall directly within that overall subject. (Thus, if the Document is in part a textbook of mathematics, a Secondary Section may not explain any mathematics.) The relationship could be a matter of historical connection with the subject or with related matters, or of legal, commercial, philosophical, ethical or political position regarding them.

The "Invariant Sections" are certain Secondary Sections whose titles are designated, as being those of Invariant Sections, in the notice that says that the Document is released

under this License. If a section does not fit the above definition of Secondary then it is not allowed to be designated as Invariant. The Document may contain zero Invariant Sections. If the Document does not identify any Invariant Sections then there are none.

The "Cover Texts" are certain short passages of text that are listed, as Front-Cover Texts or Back-Cover Texts, in the notice that says that the Document is released under this License. A Front-Cover Text may be at most 5 words, and a Back-Cover Text may be at most 25 words.

A "Transparent" copy of the Document means a machine-readable copy, represented in a format whose specification is available to the general public, that is suitable for revising the document straightforwardly with generic text editors or (for images composed of pixels) generic paint programs or (for drawings) some widely available drawing editor, and that is suitable for input to text formatters or for automatic translation to a variety of formats suitable for input to text formatters. A copy made in an otherwise Transparent file format whose markup, or absence of markup, has been arranged to thwart or discourage subsequent modification by readers is not Transparent. An image format is not Transparent if used for any substantial amount of text. A copy that is not "Transparent" is called "Opaque".

Examples of suitable formats for Transparent copies include plain ASCII without markup, Texinfo input format, LaTeX input format, SGML or XML using a publicly available DTD, and standard-conforming simple HTML, PostScript or PDF designed for human modification. Examples of transparent image formats include PNG, XCF and JPG. Opaque formats include proprietary formats that can be read and edited only by proprietary word processors, SGML or XML for which the DTD and/or processing tools are not generally available, and the machine-generated HTML, PostScript or PDF produced by some word processors for output purposes only.

The "Title Page" means, for a printed book, the title page itself, plus such following pages as are needed to hold, legibly, the material this License requires to appear in the title page. For works in formats which do not have any title page as such, "Title Page" means the text near the most prominent appearance of the work's title, preceding the beginning of the body of the text.

A section "Entitled XYZ" means a named subunit of the Document whose title either is precisely XYZ or contains XYZ in parentheses following text that translates XYZ in another language. (Here XYZ stands for a specific section name mentioned below, such as "Acknowledgements", "Dedications", "Endorsements", or "History".) To "Preserve the Title" of such a section when you modify the Document means that it remains a section "Entitled XYZ" according to this definition.

The Document may include Warranty Disclaimers next to the notice which states that this License applies to the Document. These Warranty Disclaimers are considered to be included by reference in this License, but only as regards disclaiming warranties: any other implication that these Warranty Disclaimers may have is void and has no effect on the meaning of this License.

2.VERBATIM COPYING

You may copy and distribute the Document in any medium, either commercially or noncommercially, provided that this License, the copyright notices, and the license notice

saying this License applies to the Document are reproduced in all copies, and that you add no other conditions whatsoever to those of this License. You may not use technical measures to obstruct or control the reading or further copying of the copies you make or distribute. However, you may accept compensation in exchange for copies. If you distribute a large enough number of copies you must also follow the conditions in section 3.

You may also lend copies, under the same conditions stated above, and you may publicly display copies.

3. COPYING IN QUANTITY

If you publish printed copies (or copies in media that commonly have printed covers) of the Document, numbering more than 100, and the Document's license notice requires Cover Texts, you must enclose the copies in covers that carry, clearly and legibly, all these Cover Texts: Front-Cover Texts on the front cover, and Back-Cover Texts on the back cover. Both covers must also clearly and legibly identify you as the publisher of these copies. The front cover must present the full title with all words of the title equally prominent and visible. You may add other material on the covers in addition. Copying with changes limited to the covers, as long as they preserve the title of the Document and satisfy these conditions, can be treated as verbatim copying in other respects.

If the required texts for either cover are too voluminous to fit legibly, you should put the first ones listed (as many as fit reasonably) on the actual cover, and continue the rest onto adjacent pages.

If you publish or distribute Opaque copies of the Document numbering more than 100, you must either include a machine-readable Transparent copy along with each Opaque copy, or state in or with each Opaque copy a computer-network location from which the general network-using public has access to download using public-standard network protocols a complete Transparent copy of the Document, free of added material. If you use the latter option, you must take reasonably prudent steps, when you begin distribution of Opaque copies in quantity, to ensure that this Transparent copy will remain thus accessible at the stated location until at least one year after the last time you distribute an Opaque copy (directly or through your agents or retailers) of that edition to the public.

It is requested, but not required, that you contact the authors of the Document well before redistributing any large number of copies, to give them a chance to provide you with an updated version of the Document.

4. MODIFICATIONS

You may copy and distribute a Modified Version of the Document under the conditions of sections 2 and 3 above, provided that you release the Modified Version under precisely this License, with the Modified Version filling the role of the Document, thus licensing distribution and modification of the Modified Version to whoever possesses a copy of it. In addition, you must do these things in the Modified Version:

- A. Use in the Title Page (and on the covers, if any) a title distinct from that of the Document, and from those of previous versions (which should, if there were any, be listed in the History section of the Document). You may use the same title as a previous version if the original publisher of that version gives permission.

- B. List on the Title Page, as authors, one or more persons or entities responsible for authorship of the modifications in the Modified Version, together with at least five of the principal authors of the Document (all of its principal authors, if it has fewer than five), unless they release you from this requirement.
- C. State on the Title page the name of the publisher of the Modified Version, as the publisher.
- D. Preserve all the copyright notices of the Document.
- E. Add an appropriate copyright notice for your modifications adjacent to the other copyright notices.
- F. Include, immediately after the copyright notices, a license notice giving the public permission to use the Modified Version under the terms of this License, in the form shown in the Addendum below.
- G. Preserve in that license notice the full lists of Invariant Sections and required Cover Texts given in the Document's license notice.
- H. Include an unaltered copy of this License.
- I. Preserve the section Entitled "History", Preserve its Title, and add to it an item stating at least the title, year, new authors, and publisher of the Modified Version as given on the Title Page. If there is no section Entitled "History" in the Document, create one stating the title, year, authors, and publisher of the Document as given on its Title Page, then add an item describing the Modified Version as stated in the previous sentence.
- J. Preserve the network location, if any, given in the Document for public access to a Transparent copy of the Document, and likewise the network locations given in the Document for previous versions it was based on. These may be placed in the "History" section. You may omit a network location for a work that was published at least four years before the Document itself, or if the original publisher of the version it refers to gives permission.
- K. For any section Entitled "Acknowledgements" or "Dedications", Preserve the Title of the section, and preserve in the section all the substance and tone of each of the contributor acknowledgements and/or dedications given therein.
- L. Preserve all the Invariant Sections of the Document, unaltered in their text and in their titles. Section numbers or the equivalent are not considered part of the section titles.
- M. Delete any section Entitled "Endorsements". Such a section may not be included in the Modified Version.
- N. Do not retitle any existing section to be Entitled "Endorsements" or to conflict in title with any Invariant Section.
- O. Preserve any Warranty Disclaimers.

If the Modified Version includes new front-matter sections or appendices that qualify as Secondary Sections and contain no material copied from the Document, you may at your option designate some or all of these sections as invariant. To do this, add their titles to the list of Invariant Sections in the Modified Version's license notice. These titles must be distinct from any other section titles.

You may add a section Entitled "Endorsements", provided it contains nothing but endorsements of your Modified Version by various parties--for example, statements of peer review or that the text has been approved by an organization as the authoritative definition of a standard.

You may add a passage of up to five words as a Front-Cover Text, and a passage of up to 25 words as a Back-Cover Text, to the end of the list of Cover Texts in the Modified Version.

Only one passage of Front-Cover Text and one of Back-Cover Text may be added by (or through arrangements made by) any one entity. If the Document already includes a cover text for the same cover, previously added by you or by arrangement made by the same entity you are acting on behalf of, you may not add another; but you may replace the old one, on explicit permission from the previous publisher that added the old one.

The author(s) and publisher(s) of the Document do not by this License give permission to use their names for publicity for or to assert or imply endorsement of any Modified Version.

5.COMBINING DOCUMENTS

You may combine the Document with other documents released under this License, under the terms defined in section 4 above for modified versions, provided that you include in the combination all of the Invariant Sections of all of the original documents, unmodified, and list them all as Invariant Sections of your combined work in its license notice, and that you preserve all their Warranty Disclaimers.

The combined work need only contain one copy of this License, and multiple identical Invariant Sections may be replaced with a single copy. If there are multiple Invariant Sections with the same name but different contents, make the title of each such section unique by adding at the end of it, in parentheses, the name of the original author or publisher of that section if known, or else a unique number. Make the same adjustment to the section titles in the list of Invariant Sections in the license notice of the combined work.

In the combination, you must combine any sections Entitled "History" in the various original documents, forming one section Entitled "History"; likewise combine any sections Entitled "Acknowledgements", and any sections Entitled "Dedications". You must delete all sections Entitled "Endorsements."

6.COLLECTIONS OF DOCUMENTS

You may make a collection consisting of the Document and other documents released under this License, and replace the individual copies of this License in the various documents with a single copy that is included in the collection, provided that you follow the rules of this License for verbatim copying of each of the documents in all other respects.

You may extract a single document from such a collection, and distribute it individually under this License, provided you insert a copy of this License into the extracted document, and follow this License in all other respects regarding verbatim copying of that document.

7.AGGREGATION WITH INDEPENDENT WORKS

A compilation of the Document or its derivatives with other separate and independent documents or works, in or on a volume of a storage or distribution medium, is called an "aggregate" if the copyright resulting from the compilation is not used to limit the legal rights of the compilation's users beyond what the individual works permit. When the Document is included in an aggregate, this License does not apply to the other works in the aggregate which are not themselves derivative works of the Document.

If the Cover Text requirement of section 3 is applicable to these copies of the Document, then if the Document is less than one half of the entire aggregate, the Document's Cover Texts may be placed on covers that bracket the Document within the aggregate, or the electronic equivalent of covers if the Document is in electronic form. Otherwise they must

appear on printed covers that bracket the whole aggregate.

8. TRANSLATION

Translation is considered a kind of modification, so you may distribute translations of the Document under the terms of section 4. Replacing Invariant Sections with translations requires special permission from their copyright holders, but you may include translations of some or all Invariant Sections in addition to the original versions of these Invariant Sections. You may include a translation of this License, and all the license notices in the Document, and any Warranty Disclaimers, provided that you also include the original English version of this License and the original versions of those notices and disclaimers. In case of a disagreement between the translation and the original version of this License or a notice or disclaimer, the original version will prevail.

If a section in the Document is Entitled "Acknowledgements", "Dedications", or "History", the requirement (section 4) to Preserve its Title (section 1) will typically require changing the actual title.

9. TERMINATION

You may not copy, modify, sublicense, or distribute the Document except as expressly provided for under this License. Any other attempt to copy, modify, sublicense or distribute the Document is void, and will automatically terminate your rights under this License. However, parties who have received copies, or rights, from you under this License will not have their licenses terminated so long as such parties remain in full compliance.

10. FUTURE REVISIONS OF THIS LICENSE

The Free Software Foundation may publish new, revised versions of the GNU Free Documentation License from time to time. Such new versions will be similar in spirit to the present version, but may differ in detail to address new problems or concerns. See <http://www.gnu.org/copyleft/>.

Each version of the License is given a distinguishing version number. If the Document specifies that a particular numbered version of this License "or any later version" applies to it, you have the option of following the terms and conditions either of that specified version or of any later version that has been published (not as a draft) by the Free Software Foundation. If the Document does not specify a version number of this License, you may choose any version ever published (not as a draft) by the Free Software Foundation.

How to use this License for your documents

To use this License in a document you have written, include a copy of the License in the document and put the following copyright and license notices just after the title page:

Copyright (c) YEAR YOUR NAME. Permission is granted to copy, distribute and/or modify this document under the terms of the GNU Free Documentation License, Version 1.2 or any later version published by the Free Software Foundation; with no Invariant Sections, no Front-Cover Texts, and no Back-Cover Texts. A copy of the license is included in the section entitled "GNU Free Documentation License".

If you have Invariant Sections, Front-Cover Texts and Back-Cover Texts, replace the "with...Texts." line with this:

with the Invariant Sections being LIST THEIR TITLES, with the Front-Cover Texts being LIST, and with the Back-Cover Texts being LIST.

If you have Invariant Sections without Cover Texts, or some other combination of the three, merge those two alternatives to suit the situation.

If your document contains nontrivial examples of program code, we recommend releasing these examples in parallel under your choice of free software license, such as the GNU General Public License, to permit their use in free software.
